A Short Total Synthesis of a Pseudopterosin Natural Product Featuring a Chiral Cross-Conjugated Hydrocarbon

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Cross-conjugated hydrocarbons have traditionally been considered chemical curiosities: substances exhibiting inadequate stability, with unpredictable behaviour and of dubious value. The absence of total synthesis applications involving cross-conjugated hydrocarbons attests to the fact that the synthetic community dismisses them. Herein we demonstrate that a reappraisal is warranted. We report a concise chemical synthesis of a chiral cross-conjugated hydrocarbon in enantiomerically-enriched form, and engage the substance in the shortest total synthesis of a pseudopterosin diterpene natural product. This chemical synthesis, involving four discrete acyclic precursors, is highly unorthodox compared to previous approaches, which employ starting materials with much closer structural homogeneity to the target natural product. Nevertheless, our unconventional transform-based approach has significantly lowered the step count required to access these tricyclic natural products. This work demonstrates, for the first time, that a cross-conjugated hydrocarbon can be a building block for chemical synthesis of a natural product.
The pseudopterosins (Fig. 1) represent the largest family of amphilectane diterpenes, with 31 members isolated to date, all of which are derived from one of three stereoisomeric aglycones. The remaining structural diversity arises from the nature of the sugar, the site of glycosylation, and the extent of sugar acetylation. The pseudopterosin family has been shown to exhibit a wide range of biological activities including anticancer, antimalarial and perhaps most notably, anti-inflammatory properties that exceed the potencies of existing drugs such as indomethacin.

The pseudopterosins have been the focus of a large volume of synthetic work over the past 25 years, due to their challenging structures and pronounced biological activities. To date, 14 total and formal syntheses of pseudopterosin aglycones have been published (Fig. 1). This substantial archive of outstanding synthetic contributions pinpoints the pseudopterosin family as an ideal vehicle for the development of a progressively original chemical synthesis. All previous syntheses of these chiral tricyclic hexahydro-phenalenes deploy either chiral mono-terpenes or substituted benzenes as starting materials that are converted into pseudopterosins through sequences of chain extensions and annulations. These earlier approaches are examples of structure-goal strategies: specifically, a commercially available terpene or aromatic precursor has been identified which maps onto a section of the pseudopterosin target structure. Herein we disclose the successful synthetic realisation of a transform-based strategy to a pseudopterosin: an approach which does not commence with a “mappable” commercial precursor and instead employs a powerful, triple cycloaddition sequence of a highly reactive cross-conjugated precursor to generate the natural product framework in very short order.
**Figure 1.** Previous starting material goal-based approaches to the pseudopterosins and the present, transform-based strategy, involving cross-conjugated hydrocarbon 1. *a*Synthesis of a protected derivative. *b*Formal synthesis. Step counts are reported as the longest linear sequence.

Our retrosynthetic analysis is presented in Figure 2. Viewing the catechol A-ring of pseudopterosin (−)-G-J aglycone as its 1,2-diketone tautomer 2 unveils the possibility of a Diels–Alder (DA) disconnection back to conjugated diene 3 and ethylene dione 4 as dienophile. The cyclohexene B-ring of diene 3 can be disconnected further, through a second DA transform, to provide cyclic [3]dendralene 5 and ethylene as a dienophile. A final DA disconnection of the cyclohexene C-ring of cyclic [3]dendralene 5 reveals substituted 1,1-divinylallene 1, along with 4-methyl-1,3-pentadiene 6 as dienophile.
Figure 2: Strategic bond disconnections pursued in this study.

Thus, the tricyclic framework of the natural product is exploded into four acyclic precursors through the consecutive disconnection of three pairs of covalent bonds. In the synthetic direction, issues of chemoselectivity, regioselectivity and stereoselectivity in each of the three cycloadditions would need to be overcome, in addition to the potentially problematic preparation and handling of cross-conjugated hydrocarbon 1. The presence of both E- and Z-configured propenyl-substituents in substituted divinylallene 1 confers axial chirality upon the structure, hence the possibility of a substrate-controlled stereoselective synthesis.

The preparation of chiral 1,1-divinylallene 1 in enantiomerically enriched form represented the first significant challenge of this synthesis. Our recent successful preparation of the parent 1,1-divinylallene revealed the hydrocarbon’s susceptibility toward DA polymerisation – a characteristic that, in combination with the low boiling point of the substance, dictated a somewhat lengthy synthesis\textsuperscript{19}. We ventured that chiral trimethyl analogue 1 would be both less volatile and less prone to self-immolation than its parent, and consequently set about its synthesis in a significantly more direct manner (Fig. 3). Thus, homologation of crotonaldehyde 7 into the terminal alkyne under Colvin–Hamill conditions\textsuperscript{20–22}, then deprotonation and trapping with the Weinreb amide of acetic acid furnished ketone 8 in 62% yield in one-pot, thereby avoiding the need to isolate the low boiling pentenyne intermediate. Catalytic enantioselective reduction of the ketone function of 8 under Noyori conditions\textsuperscript{23} gave propargyl secondary alcohol 9 in high enantiopurity. This compound could also be synthesised in one step from the commercially available and highly enantioenriched
propargylic alcohol 10. Thus, Sonogashira cross-coupling between (S)-3-butyne-2-ol 10 and (E)-1-bromo-1-propene 11 worked extremely well, furnishing enyne 9 in 94% yield. Alcohol 9, accessed either through the two step enantioselective synthesis or through the one step “chiral pool” pathway, was converted into the corresponding methanesulfonate derivative 12 as a prelude to the critical C–C bond forming step, which would result in the preparation of chiral cross-conjugated hydrocarbon 1, and a switch from a substance with point chirality into one with axial chirality.

![catalytic enantioselective route](image1)

![chiral pool approach](image2)

**Figure 3:** Synthesis of chiral cross-conjugated hydrocarbon 1.

After extensive experimentation, we unearthed conditions to generate hydrocarbon 1 by cross-coupling electrophile 12 with Grignard reagent 13. Our Ni(0)–catalysed Kumada cross-coupling proceeded with a high level of formal anti-

\[SN2'\]

selectivity, thereby furnishing a highly enantiomerically enriched product and, moreover, one that can be readily produced on multi-gram scale. The absolute configuration of hydrocarbon 1 was deduced through its conversion into more stable derivatives *vide infra*.

The DA cycloaddition is one of the most successful reactions in total synthesis\(^2\).\(^4\). Nevertheless, the novelty of all three contexts proposed in this synthesis caused us to be apprehensive. Of the three, we were particularly concerned about the one involving hydrocarbon 1, due to the dearth of reported examples involving axially chiral vinylallenes as dienes. We therefore modelled this process computationally using the B3LYP level of theory (see SI for details).
Scouting experiments in the laboratory uncovered the need to both replace the isobutynyl-substituent of dienophile 6 with an ester function and to include a formyl activating group at the other dienophilic carbon. Of the 43 transition structures (TSs) located for the DA addition of E-(carboxymethyl)acrolein 14-Me to the 1-E-methylbutadiene component of 1, the lowest energy TS, TS-1 predicted the formation of cycloadduct 15 and set the scene for the successful completion of the total synthesis (Fig. 4).

**Figure 4:** Total synthesis of pseudopterosin (–)-G–J aglycone by way of cross-conjugated hydrocarbon 1. CCDC1004909 contains the supplementary crystallographic data for compound 19. These data can be obtained free of charge from The Cambridge Crystallographic Centre via www.ccdc.cam.ac.uk/data_request/cif.
A control of both orientational regioselectivity and stereoselectivity in the first cycloaddition (1→15) were needed. TS-I (Fig. 4) displays significant bond-forming asynchronicity, with forming bond lengths of 2.131 and 2.661 Å (Δr = 0.53 Å). This asynchronicity confers a degree of biradicaloid character to TS-I and this is best stabilised by making the forming bond involving the allenic carbon the shorter of the two, thereby conferring pentadienyl radicaloid character to the divinylallene. The dienophile component acquires radicaloid character at the longer bond-forming carbon centre and, because the formyl group is a more potent radical stabiliser than the methoxycarbonyl group, the observed orientational preference follows. Our calculations reveal the TS with the opposite orientation to TS-I lies 4.9 kJ/mol higher in energy. The endo-CO₂Me mode of dienophile addition is favoured over the alternative exo mode by 1.1 kJ/mol and the allenic methyl group’s preference for anti over syn is 8.5 kJ/mol. Both of these preferences, together with the finding that the latter is stronger than the former, may be understood by noting that the combination of the forming bond at the allenic centre and the allenic C=C–H group in TS-I and the other TSs form a quasi-allylic system with an exo-H–C1–C2–C3 dihedral angle of 12° (cf. 0° in propene) and a C1–C3 distance of 2.86 Å (cf. 2.51 Å in propene). This quasi-allylic unit should thus be sensitive to the presence 1,3-allylic strain, which explains the allenic methyl group’s anti preference and the favoured endo-CO₂Me disposition in TS-I.

In the laboratory, the optimised first DA reaction was carried out on decagram scale and, most conveniently, in tandem with the synthesis of hydrocarbon 1. Thus, when the Kumada cross-coupling reaction was deemed complete, excess Grignard reagent 13 was quenched by the addition of methanol then commercially available dienophile 14-Et was injected into the reaction flask. The one-pot cross-coupling/DA sequence delivers adduct 15 in 61% overall yield (d.r. = 5:1:1) from alcohol 9 (a distinctly lower yield was obtained by conducting this sequence in two separate flasks) while maintaining a high level of enantiopurity over the point-to-axial-to-point chirality transfer.

The clean conversion of hydrocarbon 1 into DA adduct 15 not only reflects the unusually high reactivity of 1,1-divinylallenes as 4π cycloaddition partners but also, the low reactivity of the
s-cis diene component of 15 towards further reaction. Indeed, the lack of reactivity of the 1,3-butadiene component of 15, coupled with its similar reactivity to the diene group of adduct 16, almost undermined the synthesis. This problem was ultimately solved by halting the high-pressure reaction between triene 15 and the chosen dienophile, acrolein, at low levels of conversion, thereby minimising the amount of unwanted, acrolein double cycloaddition product. Unreacted precursor was easily re-isolated and recycled, thereby furnishing an acceptable overall yield of product 16. This reaction exhibits high regio- and stereoselectivity, with the acrolein dienophile approaching the diene from the face opposite to that in which the C3-methyl group resides. The two – now superfluous – dienophile activating groups were removed by deformylation with Wilkinson’s complex25. Ester 17 was then subjected to a one-pot selective reduction/olefination sequence to provide hydrocarbon 18 in 75% yield, thus setting the scene for the final cycloaddition.

Both strategically and conceptually, the third cycloaddition is perhaps the most interesting of the three. To our knowledge, catechol synthesis by way of a DA reaction has not been previously reported. Since ethylene dione 4 (Figure 2) has a fleeting existence under normal working conditions26, a synthetic equivalent was required. Following extensive testing involving several potential candidates, we ultimately elected to employ a synthetic equivalent of ketene and introduce the second ketone through oxidation. Thus, following a cycloaddition between hydrocarbon 18 and nitroethylene at 19 kbar and ambient temperature to give tricycle 19, a Nef reaction gave ketone 2027. Kinetic enolate formation and electrophilic oxygenation with Davis’ oxaziridine gave the resulting α-hydroxy ketone 21, which was oxidised to the pseudopterosin (−)-G–J aglycone under Swern conditions. Analytical chiral HPLC analysis of synthetic pseudopterosin G–J aglycone prepared in this manner against an authentic natural sample allowed for the assignment of absolute configuration as the (−)-G–J enantiomer (see the SI for details).

The synthesis described here should be readily amenable to the preparation of the two other naturally occurring pseudopterosin aglycones. Thus, epimerization of ester 17 followed by a repeat of the same 5 step sequence shown in Figure 4 will allow the formation of pseudopterosin K–L.
aglycone (Figure 1). Pseudopterosin A-F aglycone (Figure 1) will be accessible by simply employing either the enantiomeric Noyori catalyst or the enantiomer of the chiral pool precursor employed in this study (Figure 3).

In summary, the pursuit of a transform-based strategy has culminated in the shortest catalytic enantioselective (11 steps) and chiral pool (10 steps) total syntheses of a pseudopterosin natural product. The synthesis constructs all three rings of the tricyclic natural product via a triple DA reaction sequence commencing with an axially chiral, substituted 1,1-divinylallene. Novel and notable features of this highly unorthodox approach, which will find wider application, include (a) a new variation on the cross-coupling theme to prepare hydrocarbon 1; (b) stereoselective cycloadditions of axially chiral divinylallene 1; (c) a point-to-axial-to-point chirality maneuver with retention of enantiopurity; and (d) a novel DA reaction-based catechol synthesis. This work is perhaps the most extreme incarnation yet of the potency of the DA reaction in natural product synthesis, and one that signals the coming of age of cross-conjugated hydrocarbons in this domain.

References


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**Author contributions**

C.G.N., S.L.D., A.L.L. and M.S.S. conceived, designed and carried out the synthetic experiments. A.C.W. performed the crystallographic studies. M.N.P.-R. designed and carried out the computational study. All authors discussed and co-wrote the manuscript.

**Additional information**

Supplementary information is available in the online version of the paper. Reprints and permissions information is available online at www.nature.com/reprints. Correspondence and requests for materials should be addressed to M.S.S.

**Competing financial interests**

The authors declare no competing financial interests.
TOC graphic:

*chiral π-bond rich hydrocarbon* → *pseudopterosin (−)-G-J aglycone*

**Diagram**

- **Triple Diels-Alder sequence**
  - **7 steps**