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Hydrogen incorporation and charge balance in natural zircon

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Abstract

The water and trace element contents of natural igneous zircons were determined to constrain the mechanism of hydrogen incorporation. The low radiation-damage zircons were derived from Fe–Ti oxide gabbros from the Vema Fracture Zone (11°N, Mid-Atlantic Ridge). They contain up to 1212 ppmw H2O, 1.9 wt.% Y2O3 and 0.6 wt.% P2O5 and are generally strongly zoned. REE + Y are partially charge-balanced by P (Y, REE3+ + P5+ = Zr4+ + Si4+), but a large REE excess is present. On an atomic basis, this excess is closely approximated by the amount of H present in the zircons. We therefore conclude that H is incorporated by a charge-balance mechanism (H+ + REE3+ = Zr4+). This interpretation is consistent with FTIR data of the Vema zircons, which shows a strongly polarised main absorption band at ca. 3100 cm⁻¹, similar to experimentally grown Lu-doped hydrous zircon. The size of this 3100 cm⁻¹ band scales with H and REE contents. Apart from a small overlapping band at 3200 cm⁻¹, no other absorption bands are visible, indicating that a hydrogrossular-type exchange mechanism does not appear to be operating in these zircons. Because of charge-balanced uptake of H, P and REE in zircon, the partitioning of these elements into zircon is dependent on each of their concentrations. For instance, DREEzrc/melt increases with increasing H and P contents of the melt, whereas DHzrc/melt increases with increasing REE content but decreases with increasing P content. In addition, H–P–REE systematics of sector zoning indicate kinetic effects may play an important role. Hence, using H in zircon to determine the water content of melts is problematic, and REE partitioning studies need to take into account P and H2O contents of the melt.

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1. INTRODUCTION

The amount of water in nominally anhydrous minerals (NAMs, minerals without water in their structural formulae) has received great interest, as NAMs may hold most of the Earth’s water budget, their water content has significant effect on mineral and rock properties such as mechanical strength and electrical conductivity, and play a role in global geodynamics, such as the location of mantle discontinuities and depth of mantle melting (Smyth, 1987; Jacobson, 2006; Kohlstedt, 2006; Ohtani and Litasov, 2006; Regenauer-Lieb, 2006). In igneous rocks, the water content of NAMs may be used to reconstruct magma water contents (e.g., Kohn and Grant, 2006). The mechanism by which water (or hydroxyl) is incorporated in minerals is essential for understanding and evaluating these processes. Water uptake is often related to vacancy substitution, i.e., crystal defects (Smyth, 2006), but incorporation of other trace elements can play an essential role (e.g., Berry et al., 2005).
Zircon is a common accessory mineral and one of the most versatile geological tools, with applications ranging from U–Pb dating of igneous, metamorphic rocks and sediments, trace element contents to determine rock paragenesis and evolution, whereas oxygen, hafnium and lithium isotope studies can provide information about the genetic environment of ancient rocks (e.g., Wilde et al., 2001; Cavosie et al., 2005, 2011; Trail et al., 2007; Harrison et al., 2008; Ushikubo et al., 2008). In addition, trace elements such as REE can be used to determine the oxidation state of magmas, whereas Ti-thermometry may provide information about zircon crystallisation temperature (Ferry and Watson, 2007; Trail et al., 2007, 2012; Grimes et al., 2009; Burnham and Berry, 2012).

In contrast, the water content of zircon has received relatively little attention, perhaps due to the low abundance of zircon and, as a result, its limited effect on water budgets of Earth’s reservoirs, as well as the likelihood of the presence secondary H due to radiation damage of zircons. The limited data so far suggests low water contents of <0.01 wt.% in most natural zircons (Woodhead et al., 1991b; Bell et al., 2004). Larger amounts of water (up to 16.6%; Frondel, 1953; Coleman and Erd, 1961) can be present in metamict zircon, but this is a result of a secondary process related to destruction of the crystal structure by radiation damage due to the presence of radioactive elements such as U and Th (Aines and Rossman, 1986; Woodhead et al., 1991b; Nasdala et al., 2001). Trace elements may play a dominant role in the uptake of water in zircon, as charge imbalance in zircon (REE exceeding P) is common (Speer, 1980) and incorporation of Al, Fe, Mg and Li has been proposed to provide charge balance (Hanchar et al., 2001; Hoskin and Schaltegger, 2003). However, charge balance could also be attained by the presence of hydrogen (e.g., Woodhead et al., 1991b; Hinton et al., 2003; Trail et al., 2011), but no study exploring this relation for natural zircon yet exists.

Here we present a detailed study of water and trace element contents of natural, non-metamict zircons to constrain the mechanism of water incorporation into zircon, to determine whether water plays a role in maintaining charge balance, and to investigate if water contents of igneous zircons may be used to reconstruct the water content of host magmas.

1.1. Sample suite

Zircons were separated from Fe–Ti oxide gabbros from oceanic crust exposed at the Vema Lithospheric Section (VLS) at 11°N on the Mid-Atlantic Ridge (Auzende et al., 1988; Brunelli et al., 2006; Lissenberg et al., 2009). This suite of zircons is ideally suited for this study as they are primary magmatic zircons from a well-constrained magma source, have young ages (ca. 14 Ma) and have therefore suffered negligible radiation damage Dx of <10^18 α decay events/g (Murakami et al., 1991), and show a range of trace element zoning features, such as normal, oscillatory and sector zoning (Lissenberg et al., 2009).

The VLS is exposed along the southern wall of the transverse ridge of the Vema Fracture Zone, which rises from 5100 to 450 m below sea level. It exposes a full section of mantle peridotites, lower crustal gabbros and basaltic upper crust formed along the Mid-Atlantic Ridge (Auzende et al., 1988, 1989; Bonatti et al., 2003, 2005). The studied samples were collected by submersible craft and by dredging along a section of the VLS centred around 42°30’W and ca. 4–5000 m depth (Table 1). The gabbros commonly contain Fe–Ti oxides, indicating extensive differentiation.

Here we present data from nine zircon grains and grain fragments from four samples (Table 1). This is a subset of a larger study (Lissenberg et al., 2009; in preparation). Zircons range in size from 75 to 500 μm and generally show a subhedral to euhedral prismatic habit. Many show complex zoning patterns, but do not have xenocrystic cores and appear to have a continuous magmatic growth history. Some zircons show signs of resorption and later overgrowth (Figs. 1 and 2). These features are discussed in more detail in Section 3.1.

2. ANALYTICAL TECHNIQUES

2.1. Imaging and chemical maps (EPMA)

X-ray distribution maps were acquired with the JEOL JXA-8500F field-emission electron probe micro-analyzer at the University of Hawaii, using the following analytical conditions: 15 keV accelerating potential, 100 nA probe current and focused beam, 50 ms dwell time per pixel, a step size of 0.5–1.5 μm using a stage with linear encoders, and a resolution of individual maps of 250–550 pixels in X by 350–1000 pixels in Y. The Y-La distribution maps shown in Figs. 1 and 2 are based on combining the counts collected simultaneously on three or four spectrometers (1 PETH crystal with a Xe-sealed detector, 1 PETH crystal with gasflow detector, 1–2 PETJ with gasflow detector).

2.2. Trace elements (SIMS)

Trace elements were measured by ion microprobe (Secondary Ion Mass Spectrometry, SIMS) using a Cameca 4f monocollector instrument at the Edinburgh Ion Microprobe Facility. A beam of 16O^- was accelerated to 14.5 kV and impacted onto gold-coated samples. Two beam currents were used depending on the size of the area to be analysed: a 10 nA beam resulting in a sputtered area of ca. 20 μm diameter and a 2.5 nA beam resulting in a sputtered area of ca. 10 μm. High energy (120 ± 20 eV) secondary ions were extracted and accelerated into the mass spectrometer and counted using an electron multiplier. The instrument was operated in low mass resolution mode.

The following isotopes were measured: 30Si, 31P, 40Ca, 48Ti, 87Y, 138Ba, 139La, 146Ce, 147Pr, 143Nd, 149Sm, 151Eu, 157Gd, 159Tb, 161Dy, 165Ho, 167Er, 169Ym, 171Yb, 175Lu, 178Hf, 232Th, 238U. In addition mass 130.5 was measured to determine the electron multiplier background signal and mass 134 (^{40}Zr^{28}Si^{16}O) to allow correction of the ZrSiO interferences on Ba, La, Ce and Pr. Each analysis consisted of six cycles (sweeps through the mass spectrum). Counting times ranged from 2 s for major elements to 20 s for the lowest abundance isotopes (148Sm, 151Eu) for a total analysis time of 948 s (excluding waiting times while cycling the magnet).
Raw count rates of heavy rare earth elements were corrected for interferences from oxides of lighter isotopes (e.g., $^{139}$La, $^{140}$Ce on $^{176}$Lu) based on previously determined oxide production rates for the energy offset used (120 V). Due to the HREE-enriched pattern of zircons these corrections were generally negligible (<0.5%). Signals for $^{139}$La, $^{140}$Ce and $^{141}$Pr were corrected for ZrSiO interference based on measured $^{90}$Zr$^{2+}$SiO$_2$. This correction is 21% for La, Ce 2% and Pr 5% on zircon 91500.

Calibration was performed using NIST SRM610 using preferred values from Geomem (Jochum et al., 2011) and using its SiO$_2$ content as an internal standard. Zircon 91500 (Wiedenbeck et al., 2004) and SL1 (Maas et al., 1992) were used as secondary standards to monitor accuracy and precision. Most measured REE values for zircon 91500 are slightly lower than values from Wiedenbeck et al. (2004), which is typical for SIMS data, but are within 15% except for Yb (~22%). Rather large discrepancies are observed for Pr and La, which are 2 and 10 times too high, respectively. Both elements have very low concentrations (0.006 and 0.024 ppmw, respectively) and hence count rates are (~0.5 cps) are sensitive to, e.g., detector noise, but Nd and Sm have similar low count rates and do not show poor accuracy. Additional interferences could be ZrSiOH on La, Zr(OH)$_3$ on Pr and Nd, and CeH on Pr, but these are difficult to quantify. As Pr and La have lowest concentrations, they are most sensitive to surface contamination. Note that these elements are negligible for the total REE content of zircon. Measured Y + REE in zircon 91500 is identical to the reference value (282 ± 6 ppmw). 1s uncertainties based on 18 repeat analysis of zircon 91500 throughout the analytical session are 2% for Y, 4–6% for REE > 2 ppmw, 10–20% for REE between 0.5–2 ppmw and 20–60% for REE < 0.2 ppmw.

### 2.3. Water analysis (SIMS)

Water and selected trace elements were measured in the ion probe pits of previous trace element analyses described above, as well as in new locations in a more detailed study of one grain (S1904-03_XL5) after remounting in indium. No significant differences were observed between H$_2$O measurement in new or old pits on standard materials. The same Cameca 4f instrument was used, using a 16O$^-$ beam of 3 nA in low resolution mode. An energy window of 75 ± 20 V was used for secondary ions.

The following isotopes were measured: $^1$H, $^7$Li, $^{30}$Si, $^{31}$P, $^{40}$Ca, $^{89}$Y. A small subset of zircons was measured for additional trace elements ($^{30}$Si, $^{26}$Mg, $^{27}$Al, $^{54}$Fe, $^{92}$ZrH and $^{93}$Nb corrected for $^{92}$ZrH), but these elements were present in negligible amounts compared to REE and therefore not measured for all zircons. Each analysis consisted of ten cycles with counting times of 2 s for Si, 3 s for Ca and 5 s for all other isotopes. Phosphorus and Y were measured to allow comparison with trace element data obtained on the same analytical spots as well as to get simultaneous H, P and Y data which is important for charge-balance investigations (Sections 3 and 4). Note that even though REE are not measured during this run, previous trace element analyses on the same spots (Section 2.2) indicated an excellent correlation between Y$_2$O$_5$ and REE$_2$O$_5$ contents ($r^2 = 0.995$; Fig. 3). Therefore, Y + REE contents can be calculated from Y contents with a minimum of uncertainty.

Due to the low water content of zircons it is imperative to minimise water background signals. This was achieved in several ways: (1) samples were held for at least 48 h prior to analysis under high vacuum conditions (<10$^{-7}$ mbar in airlock, <5 × 10$^{-3}$ mbar in sample chamber); (2) analysis spots were pre-sputtered for 3 min while rastering the beam over an area ca. 20 μm wider than the analysis spot; (3) a narrow field aperture was used so that only ions from the inner 8 μm of the analysis spot are collected in the mass spectrometer; (4) only the last 5 cycles of each analysis are used to integrate the H signal (signals tend to drop about 20–40% during the first 3–4 cycles but are constant after that); (5) use of liquid N$_2$ cold finger in the sample chamber, as well as a cryogenic pump operating at 16 K; (6) whilst initial analyses were performed on zircons mounted in epoxy, grain S1904-03_XL5 used for the detailed study was remounted in indium.

The strategy detailed above resulted in relatively constant backgrounds of H$^+$ of 18–21 counts per second.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Latitude</th>
<th>Longitude</th>
<th>Depth (mbsl)</th>
<th>Lithology</th>
<th>Zircon description</th>
</tr>
</thead>
<tbody>
<tr>
<td>VE02-05$^a$</td>
<td>10.674°N</td>
<td>42.724°W</td>
<td>Not recorded</td>
<td>Undeformed, medium to coarse grained oxide gabbro</td>
<td>75–500 μm, subhedral to euhedral prismatic, commonly lacking internal structure, minor sector zoning</td>
</tr>
<tr>
<td>S1904-03$^b$</td>
<td>10.710–10.692°N</td>
<td>43.033–43.028°W</td>
<td>4900–4500</td>
<td>Strongly deformed, coarse grained oxide gabbro</td>
<td>150–700 μm, subhedral to euhedral prismatic, prominent sector and oscillatory zoning, some grains with apparent core-rim relationships and altered domains</td>
</tr>
<tr>
<td>S1923-01$^b$</td>
<td>10.703–10.687°N</td>
<td>42.575–42.594°W</td>
<td>5200–4000</td>
<td>Moderately deformed, coarse grained orthopyroxene-bearing oxide gabbro</td>
<td>60–600 μm, predominantly grain fragments, prominent sector zoning, lesser oscillatory zoning</td>
</tr>
<tr>
<td>S1923-36$^b$</td>
<td>10.703–10.687°N</td>
<td>42.575–42.594°W</td>
<td>5200–4000</td>
<td>Strongly deformed, medium to coarse grained orthopyroxene-bearing oxide gabbro</td>
<td>110–350 μm, subhedral to euhedral prismatic, prominent sector and minor oscillatory zoning, some grains lacking internal structure, rare grains with apparent core-rim relationships and altered domains</td>
</tr>
</tbody>
</table>

$^a$ VE02-05: Vemanaute cruise, RV Nadir (Auzende et al., 1988, 1989).
$^b$ S1904, S1923: Cruise S19, RV A.N. Strakhov (Brunelli et al., 2006).
(cps) for the indium-mounted zircon compared to 260 cps on experimental zircon with 380 ppmw H$_2$O. We calculated the detection limit as 15 ppmw H$_2$O based on 6× standard deviation of three H$^+$ background measurements during the analytical run. Backgrounds for the epoxy-mounted zircons were considerably higher due to outgassing of the epoxy. They averaged to 165–180 cps (based on repeat analyses of zircon 91500$^1$) compared to net H$^+$ signals on zircon of 0–400 cps. Therefore, for epoxy-mounted zircons the uncertainty of the water background (ca. 7 cps 1σ) is the largest source of uncertainty for water analysis, but amounts to less than 2% for the most water-rich zircons. Detection limit based on 6× standard deviation of the H$^+$ background is 55 ppmw for epoxy-mounted zircons.

Trace elements were calibrated using NIST SRM610 using values from Georem (Jochum et al., 2011). Calibration of water was performed using a natural obsidian glass from Lipari with 0.72 wt.% H$_2$O and 74 wt.% SiO$_2$ (unpublished FTIR and XRF data, courtesy of Prof. S. Sparks, University of Bristol) and experimental obsidian glasses from Mangan and Sisson (2000) with corrected H$_2$O values.

$^1$ Zircon 91500 is not certified anhydrous, but it consistently gave the similar low count rates for H$^+$ as experimental ‘dry zircon’ and ‘mudtank’ (Table 2).

Fig. 1. Cathodoluminescence (CL) images (left) and Y intensity maps (right) of the zircons from this study. SIMS analysis locations are marked in white or black ellipses. For zircon S1904-03_XL5 see Fig. 2.
No correction for SiO₂ content was applied as matrix effects on the ion yield of H⁺ relative to Si⁺ scale with silica contents of the samples, and therefore the same working curve can be used to calibrate rhyolites, basalts and hydrous minerals (Blundy et al., 2008; Edinburgh Ion Microprobe Facility, unpublished data).

To assess the accuracy of water contents, we analysed synthetic zircons from three separate experiments (see Electronic Supplement). The first (anhydrous) experiment was conducted in a 1 atm furnace; zircons were grown from a slowly cooled (~3°C/h) MO₃ + Li₂CO₃ flux from ~1400 to ~880 °C using a technique similar to Hanchar et al. (2001). The second was conducted in the H₂O–SiO₂–ZrO₂ system at 1650 °C 2.5 GPa, in a piston cylinder apparatus, following the procedure of Trail et al. (2011). The final experiment was similar in design, but Lu₂O₃ was added to the starting mix in order to enhance the incorporation of H into zircon through charge compensation. These three experiments yielded zircons with water contents of ~0, 200, and 750 ppmw (Table 2), respectively, as determined by FTIR (Section 2.4) following the methods described in Trail et al. (2011). The experimental zircon grains were subsequently mounted for SIMS analysis by pressing into indium, which resulted in about ten times reduction of H⁺ backgrounds. Analytical conditions were identical to those for the natural zircons, but a slightly modified isotope list was used to monitor impurities and the Lu dopant (¹H, ⁷Li, ¹⁹F, ²⁴Mg, ²⁸Si, ³¹P, ⁴⁰Ca, ⁴⁷Ti, ⁵⁴Fe, ⁸⁹Y, ⁹¹Zr, ¹⁴⁰Ce, ¹⁷⁵Lu, ¹⁷⁸Hf). CL imaging revealed sector zoning for Lu-doped and the dry zircons, whereas the undoped zircons showed patchy CL patterns (see Electronic Supplement).

In Lu-doped zircon a significant difference in Lu and H were identified between sectors, whereas in dry zircon the main difference was in P contents (Table 2). An excellent correlation was observed between H₂O by SIMS and FTIR, but the values obtained by SIMS are ca. 2 times lower.

The main uncertainty in accuracy of water contents by SIMS is the use of a rhyolite water standard. This calibration strategy gives good results for basaltic glasses (Blundy et al., 2008), which indicates that matrix-induced fractionation is successfully corrected for significantly different matrix compositions using the calibration strategy.
Table 2
Compositions of experimentally grown zircons.

<table>
<thead>
<tr>
<th>Experiment/zircon</th>
<th>Sector</th>
<th>H2O (FTIR)</th>
<th>H2O (SIMS)</th>
<th>P</th>
<th>Lu</th>
<th>Li</th>
<th>F</th>
<th>Ca</th>
<th>Mg</th>
<th>Fe</th>
<th>Y</th>
<th>Ce</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dry/z1</td>
<td>CL dark</td>
<td>n.a.</td>
<td>&lt;15</td>
<td>368</td>
<td>10</td>
<td>0.03</td>
<td>16</td>
<td>2.9</td>
<td>0.5</td>
<td>11</td>
<td>54</td>
<td>0.9</td>
</tr>
<tr>
<td>Dry/z1</td>
<td>CL bright</td>
<td>n.a.</td>
<td>20</td>
<td>143</td>
<td>16</td>
<td>0.03</td>
<td>12</td>
<td>5.3</td>
<td>0.5</td>
<td>5</td>
<td>82</td>
<td>0.3</td>
</tr>
<tr>
<td>Mudtank</td>
<td>–</td>
<td>n.a.</td>
<td>&lt;15</td>
<td>20</td>
<td>143</td>
<td>16</td>
<td>0.03</td>
<td>12</td>
<td>5</td>
<td>8</td>
<td>36</td>
<td>1.4</td>
</tr>
<tr>
<td>Muttank</td>
<td>–</td>
<td>n.a.</td>
<td>&lt;15</td>
<td>20</td>
<td>143</td>
<td>16</td>
<td>0.03</td>
<td>12</td>
<td>5</td>
<td>8</td>
<td>36</td>
<td>1.4</td>
</tr>
<tr>
<td>zw68/z1</td>
<td>Core</td>
<td>855</td>
<td>388</td>
<td>12</td>
<td>6300</td>
<td>0.02</td>
<td>14</td>
<td>5.7</td>
<td>7</td>
<td>0.8</td>
<td>0.4</td>
<td></td>
</tr>
<tr>
<td>zw73/z1</td>
<td>Core</td>
<td>855</td>
<td>388</td>
<td>13</td>
<td>6352</td>
<td>0.02</td>
<td>10</td>
<td>6.5</td>
<td>1.9</td>
<td>0.7</td>
<td>0.5</td>
<td></td>
</tr>
<tr>
<td>zw73/z1</td>
<td>CL dark</td>
<td>855</td>
<td>367</td>
<td>14</td>
<td>6026</td>
<td>0.02</td>
<td>14</td>
<td>4.9</td>
<td>1.5</td>
<td>10</td>
<td>0.7</td>
<td>0.5</td>
</tr>
<tr>
<td>zw73/z1</td>
<td>CL bright</td>
<td>n.a.</td>
<td>242</td>
<td>19</td>
<td>3778</td>
<td>0.05</td>
<td>15</td>
<td>17.8</td>
<td>6.1</td>
<td>10</td>
<td>0.5</td>
<td>0.7</td>
</tr>
<tr>
<td>zw73/z1</td>
<td>CL bright</td>
<td>n.a.</td>
<td>245</td>
<td>18</td>
<td>3900</td>
<td>0.02</td>
<td>20</td>
<td>6.7</td>
<td>2.0</td>
<td>8</td>
<td>0.4</td>
<td>0.7</td>
</tr>
<tr>
<td>zw73/z2</td>
<td>CL bright</td>
<td>750</td>
<td>354</td>
<td>32</td>
<td>6092</td>
<td>0.01</td>
<td>23</td>
<td>5.1</td>
<td>4.1</td>
<td>8</td>
<td>1.0</td>
<td>0.5</td>
</tr>
</tbody>
</table>

All concentrations in ppmw. n.a. = not analysed.

* Gem-quality zircon from Mud Tank carbonatite, Australia.

described above. Limited matrix-induced fractionation for H2O (<30%) was observed amongst various hydrous silicate minerals and silicate glasses (biotite, arfvedsonite, cordierite, clinohumite, serpentine; EIMF unpublished data), similar to findings by Ottolini et al. (2002). Although this does not imply that no matrix-induced fractionation exists for H between rhyolite and zircon, an offset of a factor two would be unexpected.

The main source of uncertainty in accuracy of water contents of zircon by FTIR is the absorption coefficient. The value published by Bell et al. (2004) of 4730 L mol⁻¹ cm⁻² is based on unpublished NRA analyses of kimberlitic zircon, but contained a math error; a corrected value of 36,241 L mol⁻¹ cm⁻² was published by Trail et al. (2011). We suspect the unpublished NRA measurement on which this value is based may have been in error. A re-evaluation of the absorption coefficient based on new NRA measurements of REE-bearing zircons is in progress.

2.4. FTIR

MicroFTIR transmitted light spectra of grain S1904-03_XL5 (resting in BaF disk after removal from epoxy puck) were collected at Bristol University on a ThermoNicolet iN10 spectrometer using a MCT/A detector and KBr beamsplitter. For E//a and E//c spectra a ZnSe polariser was used and 256 scans collected at a resolution of 4 cm⁻¹ through an aperture of 30 × 30 microns. The polished surface of the zircon was found to be very close to parallel to the c-axis as determined by EBSD (see Section 2.5). Polariised FTIR spectra //a and //c closely matched those of Woodhead et al. (1991b) and Trail et al. (2011). H2O contents were calculated using the method detailed by Trail et al. (2011) using total integrated absorbance per centimeter, calculated from polarised spectra in the two principal optical directions a and c. However, this method did not include absorption below 3000 cm⁻¹, whereas our spectra show a clear feature at ~2950 cm⁻¹. Incorporation of this feature in the water contents by extending the integrated range to 2600 cm⁻¹ increases the concentrations by about 15%. Note that combination bands around 2750 cm⁻¹ are insignificant even in the most water-poor zircons. An FTIR map of grain S1904-03_XL5 was made using an automated stage to give a 10 μm step grid. Each unpolarised spectra was acquired through a 10 × 10 μm aperture with 32 scans at 8 cm⁻¹ spectra resolution.

2.5. Electron backscatter diffraction

The crystallographic orientation relative to the polished surface of zircon grain S1904-03_XL5 was determined by electron backscatter diffraction (EBSD) using the Oxford Instruments AZtec electron backscatter diffraction (EBSD) system on a Carl Zeiss SIGMA HD VP Field Emission SEM at EMMAC, University of Edinburgh. For several points on the grain the diffraction pattern was collected and solved by the software for the orientation of zircon. Prior to EBSD analysis the sample was polished with colloidal silica in order to remove damage to the surfaces caused by earlier mechanical polishing. The e axis of zircon grain S1904-03_XL5 was oriented parallel to the polished surface within a few degrees (see Electronic Supplement).

3. RESULTS

3.1. Trace elements and zoning patterns

Trace element contents of the nine Vema zircons are presented in Table 3 and the Electronic Supplement. The zircons show a large range of Y (0.07–1.76 wt.% and ΣREE² (0.05–1.04 wt.%) contents, which mirror the zoning features visible in cathodoluminescence (CL) images (Figs. 1 and 2). Most zircons have REE-enriched cores and REE depleted rims (M9, L3, XL5, XL12, L6, L16) and usually show regions with oscillatory zoning patterns. Sector zoning is frequently observed (XL5, L5, XL2, VE1). Some zircons show CL bright, REE-poor recrystallised rims (XL5, M9), domains (L3, L16) or fracture fillings (XL5, L6, L16), probably due to deuteric alteration (cf. Grimes et al., 2009). Zircons XL5 and XL2 show CL-bright small
rectangular core domains, which are poor in REE compared to adjacent CL-dark domains. Despite the complexity of features, most zircons do not contain xenocrystic cores and appear to have undergone a continuous magmatic growth history, occasionally followed by small amounts of late-stage deuteric alteration.

The REE patterns are typical of igneous zircon (Fig. 4), with strongly depleted LREE, strong positive Ce anomalies (Ce/Ce* = 10–255) and negative Eu anomalies (Eu/Eu* = 0.2–0.6). The patterns are subparallel, i.e., despite the large range in REE concentrations the patterns are very similar. In addition to large variations in ΣREE contents, the zircons show a large range in P<sub>2</sub>O<sub>5</sub> (0.05–0.55 wt.%), Th (1–2606 ppmw) and U (8–269 ppmw). HfO<sub>2</sub> contents range from 1.06 wt.% to 2.08 wt.%.

### Table 3
Trace element compositions of two zircons from sample S1904–03 by SIMS.

<table>
<thead>
<tr>
<th>Zircon domain</th>
<th>L16</th>
<th>XL5</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Core</td>
<td>Mantle</td>
</tr>
<tr>
<td>wt.%</td>
<td></td>
<td></td>
</tr>
<tr>
<td>SiO₂</td>
<td>32.60</td>
<td>32.60</td>
</tr>
<tr>
<td>HfO₂</td>
<td>1.64</td>
<td>1.53</td>
</tr>
<tr>
<td>P₂O₅</td>
<td>0.33</td>
<td>0.26</td>
</tr>
<tr>
<td>P₂O₅&lt;sup&gt;a&lt;/sup&gt;</td>
<td>na</td>
<td>na</td>
</tr>
<tr>
<td>ppmw</td>
<td></td>
<td></td>
</tr>
<tr>
<td>H₂O&lt;sup&gt;a&lt;/sup&gt;</td>
<td>1201</td>
<td>291</td>
</tr>
<tr>
<td>Li</td>
<td>&lt;0.01</td>
<td>&lt;0.01</td>
</tr>
<tr>
<td>Mg</td>
<td>0.1</td>
<td>0.3</td>
</tr>
<tr>
<td>Al</td>
<td>&lt;0.1</td>
<td>0.4</td>
</tr>
<tr>
<td>Ca</td>
<td>0.6</td>
<td>0.9</td>
</tr>
<tr>
<td>Ti</td>
<td>11.4</td>
<td>13.0</td>
</tr>
<tr>
<td>Fe</td>
<td>5.5</td>
<td>5.7</td>
</tr>
<tr>
<td>Nb</td>
<td>9.1</td>
<td>8.6</td>
</tr>
<tr>
<td>Ba</td>
<td>0.06</td>
<td>0.09</td>
</tr>
<tr>
<td>Y&lt;sup&gt;+&lt;/sup&gt;</td>
<td>14930</td>
<td>5775</td>
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<td>Y&lt;sup&gt;a&lt;/sup&gt;</td>
<td>13845</td>
<td>5585</td>
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<tr>
<td>La</td>
<td>0.145</td>
<td>0.026</td>
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<td>Ce</td>
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<td>49.4</td>
</tr>
<tr>
<td>Pr</td>
<td>1.2</td>
<td>0.19</td>
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<td>Nd</td>
<td>22.5</td>
<td>3.8</td>
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<tr>
<td>Sm</td>
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<td>10.2</td>
</tr>
<tr>
<td>Eu</td>
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</tr>
<tr>
<td>Gd</td>
<td>233</td>
<td>73</td>
</tr>
<tr>
<td>Tb</td>
<td>97</td>
<td>33</td>
</tr>
<tr>
<td>Dy</td>
<td>1080</td>
<td>395</td>
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<td>Ho</td>
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<td>2130</td>
<td>854</td>
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<tr>
<td>Tm</td>
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<td>187</td>
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<tr>
<td>Yb</td>
<td>3688</td>
<td>1599</td>
</tr>
<tr>
<td>Lu</td>
<td>825</td>
<td>370</td>
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<td>Th</td>
<td>303</td>
<td>284</td>
</tr>
<tr>
<td>U</td>
<td>201</td>
<td>169</td>
</tr>
</tbody>
</table>

#### Sum of charge

<table>
<thead>
<tr>
<th></th>
<th>L16</th>
<th>XL5</th>
</tr>
</thead>
<tbody>
<tr>
<td>ΣY + REE</td>
<td>222</td>
<td>87</td>
</tr>
<tr>
<td>P</td>
<td>78</td>
<td>46</td>
</tr>
<tr>
<td>H</td>
<td>132</td>
<td>32</td>
</tr>
<tr>
<td>SM&lt;sup&gt;1.2.3.5+&lt;/sup&gt;</td>
<td>0</td>
<td>0</td>
</tr>
</tbody>
</table>

<sup>a</sup> Elements measured separately during a second analysis in same location with smaller beam (Section 2.3). Analysis locations are shown in Fig. 1 (L16) and Fig. 2 (XL5). Sum of charge calculated as ppmw/atomic weight × (cation charge – 4). ΣM<sup>1.2.3.5+</sup> represents cation charge of all non-tetravalent trace elements excluding H, P and REE.
REE and P, which show a strong positive correlation (Fig. 5A). However, in nearly all zircons, P$_2$O$_5$ contents are too low to completely charge-balance trivalent element contents, with P$_2$O$_5$ concentrations generally about half of those required to provide charge balance (Fig. 5A).

3.2. Zircon water contents and structural position

The water contents of zircons as measured by SIMS range from values near the detection limit (ca. 25 ppmw H$_2$O) up to 1212 ppmw H$_2$O (Tables 3 and 4). The range in water contents is large for most individual zircons and shows a clear positive correlation with P$_2$O$_5$ and REE contents (Fig. 5B and D). Recrystallised rims have lowest H and REE. To study the relationship between trace element patterns and water contents in detail, 21 analyses were performed on a single zircon, S1904-03_XL5. This zircon shows several zoning patterns, including normal, oscillatory and sector zoning (Fig. 2). It also includes a distinct core domain as well as a recrystallised rim. The range of water contents of this single zircon (<15 to 1212 ppmw) nears the range of that of the whole zircon suite (Fig. 5B) and shows distinct water-REE-P trends for different sectors and domains, which are discussed in detail in Section 4.1.

The same grain (S1904-03_XL5; Fig. 2) was also analysed by FTIR (after removal from the epoxy mount). The FTIR spectra show a highly oriented absorption band $\nu$ at 3105 cm$^{-1}$ with a smaller unresolved band at 3200 cm$^{-1}$ (Fig. 6). A hump in the tail of the 3105 cm$^{-1}$ band is observed at 2930 cm$^{-1}$ and a small additional band can be seen at 2450 cm$^{-1}$. As all spectra are similar irrespective of total absorbance, the water site is the same regardless of water content. The water contents range from 2014–2359 ppmw near the core to 93–108 ppmw in the recrystallised rim (Table 4). The size of the main 3105 cm$^{-1}$ absorption band scales with REE, H$_2$O and P$_2$O$_5$ contents measured by SIMS, although direct comparison is hampered by the different sampling volumes of the techniques (SIMS only samples the upper few $\mu$m of the grain, whereas FTIR samples the full depth, which was 79 $\mu$m) and the large fine-scale variability of trace element contents in this zircon grain (Fig. 2). A comparison between SIMS and FTIR suggests about a factor two difference in the water contents measured by the two methods (Table 4). We suspect this is mainly related to uncertainty in the FTIR absorption coefficient (see Section 2.3).

The narrow absorption bands confirm that this zircon has received a low radiation dosage and is not metamict, as metamict zircons have broad and non-polarised absorption bands stretching from 3600 to 2900 cm$^{-1}$ (Woodhead et al., 1991b; Nasdala et al., 2001; Zhang et al., 2002). This is also supported by the presence of very small but distinct combination mode peaks at 2750 cm$^{-1}$ in E//c spectra of the most H$_2$O-poor region of the zircon (cf. Woodhead et al., 1991b). The FTIR pattern is nearly identical to the ‘crystalline S.E. Asia zircon’ from Woodhead et al. (1991b). Compared to experimentally grown water-bearing zircons (this study; Trail et al., 2011), our spectra closely match those of Lu-doped zircons, which have the strongest band at 3113 cm$^{-1}$ and a smaller band at $\sim$3200 cm$^{-1}$ (Fig. 5). Other spectra, including those of zircons doped with Th, Al and Ti plus undoped experiments, do not show the main $\sim$3100 cm$^{-1}$ band (Trail et al., 2011). Vice versa, strong bands at 3384 and 3418 cm$^{-1}$ observed in zircons from kimberlites and undoped experiments (Trail et al., 2011) were not visible in our sample.

4. DISCUSSION

4.1. Water incorporation mechanism

The main substitution reaction in zircon for cations other than tetravalent cations, such as Hf, U, Th and Ti, is the xenotime reaction (Speer, 1980; Finch et al., 2001):

$$\Sigma$$REE$^{3+}$ + P$^{5+}$ = Zr$^{4+}$ + Si$^{4+}$.

(1)

However, in many natural zircons the amount of P is often insufficient to charge-balance all REE (Belousova et al.,...
2002; Hoskin and Schaltegger, 2003), including the zircons from this study (Fig. 7). Other trace elements that can potentially balance this charge deficit (such as Al, Fe, Mg, Li; Hanchar et al., 2001; Hoskin and Schaltegger, 2003) are too low in abundance to contribute <2% of the total charge compared to REE (Table 3). In contrast, H contents of zircons are close to those expected for charge balancing the P deficit (Fig. 5C) and show strong positive correlations with REE and $P_2O_5$ in Vema zircons. This indicates a possible coupled substitution mechanism for these elements. Although co-variation of elements does not necessarily imply a coupled substitution, as similar trends can be created if elements exhibit similar geochemical behaviour (e.g., comparable bulk-melt distribution coefficients), the Vema zircon data strongly suggest that REE–P–H contents are controlled by charge balance.

Hence, we propose that charge balance in the Vema zircons is achieved through a combination of the common xenotime substitution (Eq. (1)) with the following substitution mechanism (Woodhead et al., 1991b; Nasdala et al., 2001; Hinton et al., 2003; Trail et al., 2011):

\[
\Sigma\text{REE}^{3+} + H^+ = Zr^{4+}
\]  

(2)

The two substitution mechanisms (Eqs. (1) and (2)) do not operate equally in all zircons, as the H/P atomic ratio range is substantial (0.03–3.64, averaging 1.11) and increases with increasing \(\Sigma\text{REE}^3\) (Fig. 5D). In recrystallised rim and low-REE core domains, H$_2$O contents are very low and charge balance is nearly completely maintained by xenotime substitution. Sector zoning also affects REE, H and P distribution, as the REE-depleted sectors in crystal S1904-03_XL5 have higher P but lower H, REE and H + P than the REE-enriched sectors (Fig. 8). The contrasting behaviour of REE and P is particularly interesting, as typically sector zoning is regarded to increase all trace elements in one particular sector compared to another (Watson and Liang, 1995). Our data suggests that this is due to non-equilibrium effects, which be discussed further in Section 4.3.2.
FTIR data also supports a coupled substitution with REE, as the absorption spectra of Vema zircons closely match those of Lu-doped experimental zircons (Fig. 6).

Table 4
Water, P$_2$O$_5$, Y$_2$O$_3$ and $\Sigma$REE$_2$O$_3$ contents of zircon S1904-03_XL5 by SIMS and FTIR.

<table>
<thead>
<tr>
<th>Analysis</th>
<th>Domain</th>
<th>P$_2$O$_5$ wt.%</th>
<th>Y$_2$O$_3$ wt.%</th>
<th>$\Sigma$REE$_2$O$_3$ wt.%</th>
<th>H$_2$O ppm$^b$</th>
<th>H$_2$O ppm$^c$</th>
<th>H$_2$O ppm$^d$</th>
</tr>
</thead>
<tbody>
<tr>
<td>XL5-1</td>
<td>Sector 1</td>
<td>0.40</td>
<td>1.51</td>
<td>0.84</td>
<td>1212</td>
<td>1605</td>
<td>1876</td>
</tr>
<tr>
<td>XL5-2</td>
<td>Sector 1</td>
<td>0.34</td>
<td>0.99</td>
<td>0.55</td>
<td>683</td>
<td>683</td>
<td>683</td>
</tr>
<tr>
<td>XL5-3</td>
<td>Overgrowth</td>
<td>0.16</td>
<td>0.25</td>
<td>0.14</td>
<td>&lt;55</td>
<td>93</td>
<td>108</td>
</tr>
<tr>
<td>XL5-4</td>
<td>Overgrowth</td>
<td>0.07</td>
<td>0.09</td>
<td>0.05</td>
<td>&lt;55</td>
<td>93</td>
<td>108</td>
</tr>
<tr>
<td>XL5-5</td>
<td>Sector 1</td>
<td>0.47</td>
<td>1.42</td>
<td>0.79</td>
<td>939</td>
<td>2014</td>
<td>2359</td>
</tr>
<tr>
<td>XL5-6</td>
<td>Overgrowth</td>
<td>0.10</td>
<td>0.16</td>
<td>0.09</td>
<td>&lt;55</td>
<td>445</td>
<td>532</td>
</tr>
<tr>
<td>XL5-7</td>
<td>Sector 2</td>
<td>0.54</td>
<td>1.27</td>
<td>0.71</td>
<td>627</td>
<td>846</td>
<td>972</td>
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<td>XL5-8</td>
<td>Core</td>
<td>0.57</td>
<td>0.75</td>
<td>0.42</td>
<td>&lt;55</td>
<td>846</td>
<td>972</td>
</tr>
<tr>
<td>XL5-9</td>
<td>Sector 1</td>
<td>0.46</td>
<td>1.36</td>
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<td>XL5-10</td>
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<td>1.21</td>
<td>0.67</td>
<td>668</td>
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<tr>
<td>XL5-11</td>
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<td>1.04</td>
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<tr>
<td>XL5-12</td>
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<tr>
<td>XL5-13</td>
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<td>1211</td>
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<tr>
<td>XL5-14</td>
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<td>0.70</td>
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<td>251</td>
<td>845</td>
<td>1013</td>
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<tr>
<td>XL5-15</td>
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<tr>
<td>XL5-16</td>
<td>Sector 2</td>
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<td>0.77</td>
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<td>333</td>
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<td>XL5-17</td>
<td>Sector 1</td>
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<td>0.60</td>
<td>0.34</td>
<td>328</td>
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<tr>
<td>XL5-20$^e$</td>
<td>Sector 2</td>
<td>0.38</td>
<td>1.30</td>
<td>0.72</td>
<td>609</td>
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<tr>
<td>XL5-32$^e$</td>
<td>Sector 1</td>
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<td>1.44</td>
<td>0.80</td>
<td>815</td>
<td>815</td>
<td>815</td>
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<tr>
<td>XL5-33$^e$</td>
<td>Sector 1</td>
<td>0.36</td>
<td>0.92</td>
<td>0.51</td>
<td>507</td>
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<tr>
<td>XL5-34$^e$</td>
<td>Overgrowth</td>
<td>0.09</td>
<td>0.11</td>
<td>0.06</td>
<td>&lt;15</td>
<td>445</td>
<td>532</td>
</tr>
</tbody>
</table>

* Calculated from Y$_2$O$_3$ contents following the equation in Fig. 2.

$^b$ SIMS data.

$^c$ FTIR data integrating 3550–3000 cm$^{-1}$.

$^d$ FTIR data integrating 3550–2600 cm$^{-1}$.

$^e$ Grain remounted in indium with lower H$_2$O detection limit of 15 ppmw. Analysis locations are shown in Fig. 2. Analyses XL5_1 to XL5_4 match those listed for zircon XL5 in Table 2.

Fig. 6. Zircon FTIR spectra of various analysis spots on grain S1904-03_XL5 and for experimentally grown zircons. Spectra have been vertically displaced for clarity. Analysis locations for grain S1904-03_XL5 can be found in Fig. 2. Calculated water content of each spot is indicated for each spectrum (see Section 2.4 for details of calculation). Experimental zircons are ‘undoped’ and ‘Lu-doped’ hydrous zircons (Table A1).

FTIR data also supports a coupled substitution with REE, as the absorption spectra of Vema zircons closely match those of Lu-doped experimental zircons (Fig. 6). Data from undoped experimental zircons indicates that only up to 200 ppmw H$_2$O can enter the zircon structure (at 1650 ˚C, 2.5 GPa; this study, Trail et al., 2011), whereas doping with trace elements, including REE, strongly increased the zircon water contents, providing clear evidence for a coupled substitution mechanism.
One intriguing aspect of the Vema zircon FTIR data is the absence of commonly reported OH bands at ~3400 cm\(^{-1}\). Woodhead et al. (1991b) argued that sharp peaks at 3420 cm\(^{-1}\) (E//c) and 3385 cm\(^{-1}\) (E.Lc) mark the OH site in zircon, and that the absence of this peak meant that the zircon was anhydrous. Our data disprove this, as the Vema zircons lack the 3400 cm\(^{-1}\) peak despite having high water contents. Trail et al. (2011) speculated that the 3400 cm\(^{-1}\) peak represented an OH site related to a Zr vacancy and suggested that the 3200 cm\(^{-1}\) band is related to a hydrogrossular substitution (Si vacancy: \(\text{SiO}_4^{3-} \rightarrow [\text{OH}]_4\)). As the 3200 cm\(^{-1}\) band is clearly present in our spectra, and no vacancies are created in the two charge-balancing exchange reactions (Eqs. (1) and (2)), we infer that this band is probably not vacancy related. This is supported by a recent molecular dynamics study which calculated the most stable configurations of OH incorporation and associated OH bond lengths, from which FTIR absorption bands can be predicted (Botis et al., 2013). The authors concluded that the 3100 cm\(^{-1}\) and 3220–3250 cm\(^{-1}\) peaks are related to substitution mechanism Eq. (2), where the hydrogen ion can be either in a position close to the M\(^{4+}\) cation (3220–3250 cm\(^{-1}\) band) or a far position (3100 cm\(^{-1}\) band). Both peaks are observed in the Vema zircons (Fig. 6). The study also assigned the (partial) hydrogrossular substitution to a 3420 cm\(^{-1}\) band. As this mechanism involves a Si vacancy, it seems reasonable to assume that it most likely occurs in rocks with low silica activities such as kimberlites, as is the case for, e.g., olivine (Berry et al., 2005). The available literature appears to confirm this (Woodhead et al., 1991b; Bell et al., 2004), which may explain the lack of this mode of OH incorporation in our zircons, which are from rocks with high silica activities. In addition, zircons from kimberlites commonly show an excess P over REE (Belousova et al., 2002), so no H is needed for charge balance (Fig. 7). We note, however, that the ~3400 cm\(^{-1}\) band is nearly absent in Lu-doped experiments from Trail et al. (2011), but does appear in non-doped experiments, despite similar starting materials and experimental conditions. This suggest that the hydrogrossular substitution may be suppressed in the presence of REE + H substitution.

4.2. Maximum water and REE contents of non-metamict zircons

The maximum amount of water that can be incorporated in zircons for a given pressure, temperature and water activity in the melt has yet to be determined, but is perceived to be very limited, e.g., 0.01 wt.% by Woodhead et al. (1991b). This in contrast to metamict zircon, that can contain larger amounts of structural water (up to 0.04 wt.%, Woodhead et al., 1991a) plus significant amounts of secondary water (Nasdala et al., 2001; Zhang et al., 2002). Here we report maximum water contents of 0.12 wt.% \(\text{H}_2\text{O}\) in REE-rich crystalline zircons, indicating that such water contents cannot simply be used to infer a metamict state of the zircon without knowledge of its REE contents or IR spectral evidence of the OH site (Cliff et al., 2013). This is supported by experimental data by Trail et al. (2011), who showed that very high \(\text{H}_2\text{O}\) contents (>0.17 wt.%) can be accommodated by REE-rich crystalline zircon, perhaps aided by the high \(P\)-\(T\) conditions of the experiments (1650 °C, 2.5 GPa). The upper water limit may be determined by increased lattice strain at the Zr site during incorporation of REE, as REE are larger than Zr (\(^{88}\text{Zr}^{4+}\) 0.84 Å, \(^{106}\text{Y}^{3+}\) 1.02 Å). During the xenotime substitution this is partially balanced by substitution of the smaller P cation on the Si site (\(^{40}\text{Si}^{4+}\) 0.26 Å vs. \(^{44}\text{P}^{5+}\) 0.17 Å) (Finch et al., 2001), but the incorporation of H will have no such effect and results in increased lattice strain. One would therefore expect for H/P ratios to decrease with increasing H and REE contents, but we observe the opposite although the data is rather scattered (Fig. 5).

Similarly, Hoskin and Schaltegger (2003) state that zircon analyses with significantly more than 1 wt.% \(\text{Y} + \Sigma\text{REE}\) must be the result of alteration or contamination by accidental analysis of inclusions. However, the \(\text{Y} + \Sigma\text{REE}\) content of Vema zircons reaches a maximum of 2.94 wt.%, an amount which is completely charge-balanced by \(\text{P}_2\text{O}_5\) and \(\text{H}_2\text{O}\) (0.55 wt.% and 1201 ppmw, respectively). Even higher \(\text{REE} + \text{Y}\) contents (up to ca. 5 wt.%) were reported by Grimes et al. (2009), but no H contents were reported by the authors. If REE in zircon were due to
accidental analysis of xenotime inclusions, the zircons would plot on the 1:1 line in Fig. 7, whereas in fact most plot below this line. Therefore, although the upper limit of REE solubility in zircon is poorly constrained, our results demonstrate that it is strongly dependent on P and H activities in the system and extends to much higher values than has been previously suggested.

4.3. Implications for the use of zircon to estimate melt water contents and zircon REE partitioning

4.3.1. Equilibrium partitioning

An important outstanding issue is whether H in zircon can be used to constrain the water activity of melt (e.g., Trail et al., 2011). As H uptake in zircon is in a charge-balance mechanism with REE and P, the contents of the latter need to be taken into account. The concentration of a trace element i in the melt can be estimated from its concentration c_i in a mineral and an experimentally determined distribution coefficient D_a, assuming other elements exert no influence on the distribution of element i. However, if element i is part of a coupled substitution with element j, the concentration c_j of the latter must be taken into account. For a reaction melt + mineral = mineral, we can write an equilibrium equation:

\[
\frac{c_{\text{min}}^i \times c_{\text{min}}^j}{c_{\text{melt}}^i \times c_{\text{melt}}^j} = K_D
\]

Thus, the concentration, or more accurately, the activity of H in zircon, as incorporated following Eq. (2), depends on REE activity of the melt as follows:

\[
\text{(H)}_{\text{zrc}} = \frac{K_{D/1}(\text{H})_{\text{melt}}(\text{REE})_{\text{melt}}}{(\text{REE})_{\text{zrc}}}
\]

(4)

assuming constant activities of Zr in zircon and the melt. The latter assumption is valid when zircon saturation is reached. Note that to satisfy charge-balance constraints, [H]_{zrc} = [REE]_{zrc}. A similar equation can be written for P and REE in the xenotime reaction Eq. (1), ignoring the presence of water:

\[
\text{(P)}_{\text{zrc}} = \frac{K_{D/2}(\text{P})_{\text{melt}}(\text{REE})_{\text{melt}}}{(\text{REE})_{\text{zrc}}}
\]

(5)

assuming constant activities of Zr and Si in melt and zircon. If both H and P are present, they compete for REE to obtain charge balance in zircon and the concentrations of all three elements in zircon are mutually dependent on their concentrations in the melt. The ratio of H and P in zircon is equal to K_{D1}/K_{D2}.

As Eqs. (4) and (5) are essentially the same, the combined solubility of H + P can be expressed, as a first approximation, as:

\[
\text{(H + P)}_{\text{zrc}} = \frac{K_{D1}(\text{H + P})_{\text{melt}}(\text{REE})_{\text{melt}}}{(\text{REE})_{\text{zrc}}}
\]

(6)

where K_{D1} = f_{H1} \times K_{D1} + (1 - f_{H1}) \times K_{D2} with f_{H1} = (H)/(H + P)_{melt} and the ratio of H and P in zircon is proportional to that in the melt:

\[
\frac{(\text{H})_{\text{zrc}}}{(\text{P})_{\text{zrc}}} = k \frac{(\text{H})_{\text{melt}}}{(\text{P})_{\text{melt}}}
\]

(7)

This set of equations allows us to calculate the concentration of H, P and REE in zircon based on their concentrations in the melt, if K_{D1} and K_{D2} are known and assuming activity factors of one. An important consequence of the above formulation is that the same zircon composition can be derived from different melt compositions, so for a given water content of zircon, the water content of the melt can only be constrained if either REE or P content of the melt can be determined independently. Note that although our treatment differs in detail from that by Wood and Blundy (2001), as their model is based on the composition of the crystal and not the melt, they also predicted matrix-dependent D values for charge-coupled substitutions.

An additional consideration is the activity of H in the melt, as water dissolves as hydroxyl (OH^−) and molecular water (H₂O*) in siliceous melts (Stolper, 1982; Zhang, 1999). The relative proportions of the two water species depend on the total water content of the melt; at low water contents most H₂O dissolves as OH^−, but H₂O* becomes increasingly important with increasing water contents and above about 3–5 wt.% H₂O, most water dissolves as H₂O* (Ihinger et al., 1999; De Hoog et al., 2009). As water is incorporated into zircon as hydroxyl, partitioning of H between zircons and melt could be affected by this speciation change.

The interdependence of P, H and REE has important implications for REE zircon-melt partitioning studies. It has been noted that D values determined from natural samples and experimental studies vary by at least two orders of magnitude (Hancher and van Westrenen, 2007). The authors indicate several possible reasons, including zoning of crystals, analytical difficulties, presence of REE-rich accessory phases and melt structure. However, as shown in Eqs. (4) and (5), the Nernst equation for partition coefficients does not apply when charge balance is needed to incorporate a certain trace element. Only if charge balance is maintained by exchange for major elements, the concentrations of which can be assumed approximately constant in many systems (e.g., REE in augite through the reaction REE + Al = Ca + Si), can a charge-balance coupled substitution be approximated by a constant D value. However, in the case of REE in zircon, the concentrations of H₂O and P₂O₅ vary widely between different natural systems and change significantly during magma evolution, and therefore so will D values. For instance, D_{REE}^{Zr/melt} increases with increasing H and P contents of the melt, whereas D_{REE}^{H} increases with increasing REE content but decreases with increasing P content. This makes inferring melt water contents from H in zircon particularly challenging.

4.3.2. Kinetic effects

The solution model presented in Section 4.3.1 assumes equilibrium partitioning of H, P and REE during zircon growth. However, all zircons from this study show oscillatory zoning and/or sector zoning (Figs. 1 and 2). Oscillatory zoning may be the result of external forcing,
e.g., magma mixing, convection or volatile release, or intrinsic crystal growth processes related to varying diffusivities of major and trace elements in the boundary layer surrounding the growing crystal (Hoskin, 2000; Fowler et al., 2002). As diffusivities of most elements in zircon are low, oscillatory zoning will be preserved during continued crystal growth (Watson and Liang, 1995).

Sector zoning has been related to different crystal faces having different bonding energies and therefore differential uptake of trace elements from the melt (Dowty, 1976). Whether the occurrence of sector zoning is related to crystal growth rate is disputed (Watson and Liang, 1995), but growth rates of different sectors vary and therefore any crystal growth-rate related effects such as boundary layer diffusion will be amplified.

In Vema zircon S1904-03_XL5, H and REE are highest in the slow growing sector (CL dark, //c-axis), whereas P is highest in the fast growing sector (Fig. 8). Although P contents of the melts are unknown, they were likely to be higher than P contents of the zircons (P2O5 = 0.05–0.55 wt.%) as the melts were apatite saturated. Therefore, P was present in excess near the growing zircon crystal and needed to diffuse away during zircon growth. The same applies to H, as it must have been in excess in the melt, but REE (with the exception of LREE) are enriched in zircon relative to the melt and thus needed to diffuse towards the growing zircon. As P is one of the slowest diffusing elements in silicate melts (Harrison and Watson, 1984), during rapid crystal growth P will be unable to diffuse away from the growing crystal surface and become trapped, leading to excess P in zircon. Due to the excess P, less H is needed to charge balance REE and H/P will decrease in the faster growing sector, as is indeed observed (Fig. 8). One would expect the REE concentration to increase to compensate for the increased P contents, but in reality REE decrease in the P-rich sectors (Fig. 8). This suggests that perhaps REE incorporation is also diffusion limited.

It is difficult to constrain if the H–P–REE systematics observed for sector zoning also characterise oscillatory zoning, as the oscillations are too narrow to be measured individually as well as superimposed on a strong normal zoning pattern. The overall fractionation trend of the zircons is a simultaneous decrease of H, P and REE, which differs from the sector zoning and is consistent with the equilibrium fractionation model.

4.3.3. Implications for REE patterns

The shape of REE patterns is probably relatively unaffected by the fractionation effects indicated above and may therefore still provide useful genetic information about the source rock (e.g., Belousova et al., 2002). Important potential exceptions are Ce and Eu anomalies. Cerium anomalies are common in zircon due to the preferential uptake of Ce3+ over Ce4+, both of which occur in siliceous melts in relative proportions that depend on the melt oxidation state (Hinton and Upton, 1991). As Ce4+ exchanges directly for Zr4+, its partitioning will be independent of H and contents, in contrast to Ce3+. Thus, Ce anomalies will be different for zircons from melts with identical oxidation states but different H or P contents. For instance, using the equilibrium partitioning model, increasing both H and P in the melt by a factor two would result in an increase in La and Pr in zircon by a factor two. In contrast, the increase in Ce would be negligible as it is dominated by Ce4+, which is not dependent on H or P contents. In a zircon with Ce/Ce* = 20 initially, Ce/Ce* would therefore decrease to 10.5, equivalent to a decrease in apparent fO2 by 2.5 log units according the model by Trail et al. (2012). This is similar to experimental observations by Burnham and Berry (2012), who noticed that in experiments with high oxidation states and therefore high Ce4+ activity, Ce differed little between different sectors, whereas all other REE were clearly affected. This limits the applicability of experimental Ce partitioning data for estimating melt oxidation state (Burnham and Berry, 2012; Trail et al., 2012) to systems with broadly similar H and P contents. In contrast, Eu anomalies are probably not affected, as in zircon it is a reflection of the Eu anomaly of the melt and not a result of preferential incorporation of one of the species into zircon (Hoskin and Schaltegger, 2003) and Eu4− and Eu3+ are likely to be similarly affected by charge balance effects on trace element partitioning.

It is worth noting that Eq. (4) does not apply to water incorporated in zircon by other substitution mechanisms. For example, the average H2O solubility in zircon is found to be 188 ppmw in undoped zircons (Trail et al., 2011), which is the result of hydrogrossular substitution. Potentially, H in zircon may be used to constrain water contents in REE-poor melts with low silica activities, but this will need additional experimental confirmation. However, H2O contents of REE-bearing siliceous melts cannot be accurately reconstructed from H in zircon without knowledge of P and REE contents of the melt. Further work is needed to constrain the interdependent partition coefficients of H, P and REE between zircon and melt.

5. CONCLUSIONS

(a) Water contents as measured by SIMS in zircons from Fe-Ti gabbros from the Vema Lithospheric Section (Mid-Atlantic Ridge) range from <15 to 1212 ppmw and correlate strongly with REE and P contents. FTIR absorption spectra of these zircons are very similar to those of experimentally grown Lu-doped hydrous zircons, showing a main absorption band at 3100 cm−1.

(b) Water is incorporated following the nominal reaction H+ + REE3+ = Zr4+ and therefore competes with P for REE, as P is incorporated following the xenotime reaction: P5+ + REE3+ = Zr4+ + Si4+. Other trace elements are too low in concentration to play a role in the charge balance.

(c) Equilibrium partitioning of charge-coupled trace elements depends on the activity in the melt of all elements taking part in the exchange reaction. Thus water uptake in zircon depends on not only on H but also P and REE contents of the melt. This makes estimating water contents in the melt from H in zircon very challenging.
(d) Most zircons show sector zoning. The fastest growing sector is enriched in P but depleted in H and REE compared to the slower growing sector. As P is depleted in zircon relative to the melt, REE are enriched, this suggests that diffusion to (for REE) and from (for P) the boundary layer limits the uptake of these elements in zircon.

(e) REE partitioning in zircon is dependent on H and P contents of the melt. As Ce in zircon occurs primarily as Ce\(^{4+}\) and therefore exchanges directly for Zr (and \(^{16}O\)) by ERC Advanced Grant “CRITMAG.” We thank Enrico Bonatti, Daniele Brunelli and the IFREMER Lithothèque for providing samples. Laurence Coogan and Jon Blundy are thanked for discussions about REE partitioning into zircon and Nicola Cayzer for EBSD analysis and interpretation. Jed Mosenfelder and an anonymous reviewer are thanked for constructive comments and Anders Meibom for editorial handling.

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**APPENDIX A. SUPPLEMENTARY DATA**

Supplementary data associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.gca.2014.06.033.

**REFERENCES**


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