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A Computationally Designed Binding Mode Flip Leads to a Novel Class of Potent tri-vector Cyclophilin inhibitors

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Cyclophilins (Cyps) are a major family of drug targets that are challenging to prosecute with small molecules because the shallow nature and high degree of conservation of the active site across human isofoms offers limited opportunities for potent and selective inhibition. Herein we report a computational approach based on molecular dynamics simulations and free energy calculations was combined with biophysical assays and X-ray crystallography to explore a flip in the binding mode of a reported urea-based Cyp inhibitor. This approach enabled access to a distal pocket that is poorly conserved among key Cyp isoforms, and led to the discovery of a new family of sub-micromolar cell-active inhibitors that offer unprecedented opportunities for the development of next-generation drug therapies based on Cyp inhibition. The computational approach is applicable to a broad range of organic functional groups and could prove widely enabling in molecular design.

Introduction

In the absence of innovative treatments, the societal burden of neurodegenerative disorders is projected to grow substantially over the next few decades worldwide. A growing body of evidence points to the central role of the opening of the mitochondrial permeability transition pore (mPTP) in neurodegenerative cell death. The enzyme cyclophilin D (CypD) is an essential regulator of the mPTP opening, and blockage of CypD function is neuroprotective in models of Alzheimer’s and Parkinson’s diseases. Besides neurodegenerative disorders, CypD inhibition is protective in animal models of acute kidney injury, a condition that accounts for a fifth of emergency hospitalizations in the USA. Other related cyclophilin isoforms CypA and CypB are overexpressed in malignant cancers, and CypB inhibition has potent antiviral and anti-inflammatory effects. Cyp inhibition may also protect the liver from fibrosis subsequent to non-alcoholic steatohepatitis (NASH), a common indicator for liver transplantation after chronic hepatitis C.

The majority of existing Cyp inhibitors are derived from the natural products Cyclosporin A (CsA) and Sanglifehrin A. These scaffolds achieve potent inhibition but are complex to synthesize, lack sub-type selectivity, show undesirable side-effects, and have poor CNS activity. Extensive efforts have been pursued to simplify these scaffolds into more synthetically tractable variants. There is also considerable interest in developing small molecule Cyp inhibitors that may be more easily optimized for use as therapeutics. However, there are few reports of well-characterized small molecules that bind strongly to Cyps. Current small molecule design strategies typically feature linear (thio)ureas or amides as scaffolds. Compound 1 provides an example that exploits interactions with residues in the highly conserved Pro and Abu pockets (Fig 1A). Fig 1A also highlights a less conserved distal three o’clock pocket that is thought to offer prospects for potency and selectivity improvements, but to date it is unclear how this pocket can be engaged using current scaffolds.

Herein we report a multidisciplinary approach to redesigning linear Cyp inhibitor scaffolds into the first class of tri-vector inhibitors that simultaneously target the Pro, Abu and three o’clock pockets.
**Results and discussion**

**Molecular dynamics simulations suggest a new ligand optimisation strategy**

The protein X-ray structure of 1 in complex with CypA suggests that both urea nitrogen atoms are hydrogen-bonded to the backbone oxygen of Asn102. However, molecular dynamics (MD) simulations revealed that the nitrogen atom distal to the ester moiety in 1 is only weakly interacting with Asn102 and readily moves away from this residue (Fig 1B, 1C & Fig S1), suggesting that alkylation of that nitrogen to introduce a new vector in the scaffold could be tolerated. Ab-initio calculations on model ureas suggest a modest energetic preference for the ZZ urea conformer. This led to the hypothesis that a suitably chosen R group could stabilize an alkylated urea in an EZ conformer able to adopt a novel type-II binding mode that would enable access to the 3 o’clock pocket (Fig 1D).

**A general synthetic route to tri-vector Cyclophilin ligands**

The tri-vector design hypothesis was pursued by synthesis of a series of alkylated urea ester ligands. The majority of the compounds were prepared according to Scheme 1 (steps a-d, see ESL for details). Substituted amino derivatives were typically prepared from 4-nitrophenyl derivatives by reductive amination, or nucleophilic substitution. The urea moiety was prepared by condensation of the resulting amines with ethyl isocyanatoacetate, and the desired products were obtained by subsequent reduction of the nitro group.

**A free-energy perturbation (FEP) protocol guides the selection of substituents**

MD simulations cannot readily sample the desired type-I to type-II binding mode flip because rotational barriers of ca. 15 kcal.mol\(^{-1}\) separate ZZ and EZ conformers. Free Energy Perturbation (FEP) calculation methods have shown potential for optimization of ligand potency and selectivity. Here the selection of potential R groups was guided by a novel FEP protocol designed to compute the energetics of this binding mode flip (Fig 2A). The setup, execution and analysis of the FEP calculations were carried out with the software FESetup, and freenrg workflows. The selection of R groups for computational and experimental evaluation was driven by the following considerations: lack of obvious steric or electrostatic mismatches, polar or non-polar interactions with 3 o’clock pocket residues, ease of synthesis, diversity of structure–activity relationships, and validation of FEP-calculated energetics.
Compound 1 was predicted to favour a type-I binding mode by ca. 1 kcal mol\(^{-1}\), in line with the preferences observed in X-ray structures (Fig 2A). Non-polar substitutions (2, 3, 4, 5, 7) were predicted to worsen free energies of binding relative to 1 and maintain preference for a type-I binding mode. By contrast, nitrogen-rich 5-membered rings (8, 9, 10) were predicted to show a similar or slight preference for type-II binding mode.

Ligand binding to CypA was characterized by isothermal titration calorimetry (ITC) measurements (Fig 2B). Substitutions with small linear alkyl groups (2, 4) were tolerated but weakened binding, whereas branched alkyls (3) were inactive, in line with the FEP calculations. Alkynes, cyano and cyclopropyl derivatives (5, 6, 7) were computed to bind with potentials comparable to linear alkyl groups, but turned out to be very weak binders in the ITC assay. Nitrogen-rich five-membered rings (8, 9, 10) showed more favourable binding constants, in line with the FEP calculations. Piperidines 11 and 12 had not been evaluated by FEP calculations and turned out to be inactive.

Evidence of binding mode preferences was sought by soaking CypA crystals with compounds 2-12. Seven X-ray structures of CypA complexes were determined. Compounds 2, 4, 7 were found to adopt a type-I binding mode in line with the FEP calculations (Fig 2C and ESI). Compound 5 was observed to adopt dual occupancy, whereas the FEP calculations anticipated a type-I binding mode. Compound 8 was observed to adopt dual occupancy, whereas 9 and 10 were only observed in a type-II binding mode in the X-ray refined structures. The FEP calculations suggest both binding modes are feasible for these three compounds. In 8-10 the type-II binding mode is stabilized by hydrogen-bonding interactions between nitrogen atoms in the five-membered rings and His54. This binding mode flip from type-I to type-II is accompanied by a slight rearrangement in the orientation of the urea carbonyl to maintain optimal placement of the aniline ring in the Abu pocket. Strikingly the 5-membered ring methyl groups in 8 and 9 point into the 3 o’clock pocket of CypA and provide a clear vector for further derivatization (Fig 2D).

The tri-vector design generalises to other ligand families and Cyp isoforms.

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**Fig. 2** A) FEP-calculated binding energetics. The figures are relative to 1 in a type-I binding mode and were obtained from automated analysis of FEP maps (Fig S2). A partial map depicted below the table illustrates how molecule Int04 is used to connect other compounds in a type-I or II binding mode. B) ITC-derived dissociation constants and observed binding mode in X-ray crystallography derived CypA structures. Uncertainties on \(K_d\) values from fit to a one-site binding model. Partial occupancies are given for compounds refined in multiple binding modes. n.b.: no binding observed at the maximum concentration tested. n.d.: not determined. C) X-ray crystal structure of CypA:2 depicting a type-I binding mode. D) X-ray crystal structure of CypA: 9 depicting a type-II binding mode. Fo − Fc electron density omit maps are shown as a green mesh at 2.5\(\sigma\) contour (see ESI for details).
With the binding mode flip hypothesis validated for the urea ester series in CypA, attention turned to transferability of the strategy to other ligand families and Cyp isoforms. FEP calculations suggested that a type-II binding mode preference was maintained for nitrogen-rich heterocycles in the context of a urea arylpyrrolidine scaffold for both CypA and CypD isoforms (Figs S3-S5).\textsuperscript{21} Thus alkylated variants 13-16 were prepared from 4-nitrophenyl ester intermediates (Scheme 1, steps e and d). Compounds 13-15 feature similar R groups to 5, 8 and 9, whereas 16 was prepared to assess viability of further extension into the three o’clock pocket via an appendage of an ester group to a triazole. The X-ray crystallography derived structure of 15 in complex with CypA confirmed that the type-II binding mode has been maintained (Fig 3A). Thus 13-16 were assayed by surface plasmon resonance (SPR) against CypA, CypB and CypD isoforms. The compounds bind to all three isoforms with $K_d$ values in the low $\mu$M to mid nM range (Fig 3B). Compound 15 binds particularly well to CypD, with a potency (ca. 70 nM) approaching that of CsA (ca. 20 nM). For comparison, the unaltered (R=H) urea has been reported to inhibit CypD with an $IC_{50}$ of 1.1±0.2 $\mu$M.\textsuperscript{15} Although 13-15 only access the entrance of the three o’clock pocket, the SPR assay suggests this is sufficient to endow a degree of isoform selectivity. Compound 16 binds slightly worse to CypA and CypB but modestly better to CypD with respect to 14. Thus further lead optimization efforts to tune desired selectivity profiles appear auspicious.

Tri-vector ligands show improved efficacy and decreased toxicity over Cyclosporine in cell assays

Additional evidence for the potential of this tri-vector design was sought by evaluating 13-15 in cellular assays. CypA has previously been shown to be necessary for the prolactin-induced activation of Janus activated kinase 2 in human breast cancer cells, with CsA treatment causing inhibition of growth of the triple negative MDA-MB-231 breast-cancer cell line.\textsuperscript{44} Compounds 13 and 15 showed evidence of dose-dependent growth inhibition at low micromolar concentration resulting in $GI_{50}$ values similar (13) or two-fold better (15) than those measured for CsA. By contrast 14 was inactive (Fig 4A-B). Thus alkylation of the urea moiety does not appear to prevent cell penetration. Negligible cell death was observed for compounds 13 and 15 suggesting low toxicity in comparison with CsA. This indicates that in MDA-MB-231_NLG cells growth inhibition by compounds 13 and 15 is due to reduced proliferation, whereas growth inhibition by CsA is also a consequence of cell death (Fig 4C). Additionally 13-15 showed no growth inhibition or induction of cell death in the non-tumorigenic
fibroblast IMR-90 cell line. By contrast CsA induced significant growth inhibition and cell death (Fig 4D).

Thus, further evaluation of the present tri-vector Cyp inhibitors in a range of in vitro and in vivo disease models implicating Cyps seems to be warranted, particularly in instances where CsA toxicity and off-target effects have been a cause for concern. Alternatives to the aniline moiety in 15 to alleviate toxicity concerns have been suggested elsewhere.\textsuperscript{21, 45} Immunosuppression via inhibition of calcineurin phosphatase activity is a well-known feature of CsA. The tri-vector Cyp inhibitors are unlikely to possess this capability given that the parent urea-arylpyrrolidine compounds they are derived from lack calcineurin inhibition properties.\textsuperscript{21}

Conclusions

In summary, this work has validated a strategy for the redesign of existing cyclophilin inhibitors into a novel class of inhibitors that can simultaneously target three pockets on the surfaces of cyclophilins. This breakthrough has led to significantly more potent cyclophilin inhibitors and also opens up new horizons to optimise isomor-selectivity. Nanomolar acylurea-based cyclophilin inhibitors have been reported previously,\textsuperscript{46} but subsequent work was unable to reproduce these findings or to detect evidence of binding to cyclophilins for these molecules.\textsuperscript{21} Members of the tri-vector class of cyclophilin inhibitors (13, 15) show improved efficacy and reduced toxicity in cell assays over the drug Cyclosporine A. They are thus already attractive leads for new drug therapies based on Cyp inhibition. To our knowledge this is the first example of the validation of an FEP methodology to computationally design a large scale binding mode flip in a protein-ligand complex. The approach is readily applicable to a broad range of commonly encountered organic functional groups that exhibit E/Z conformational equilibria. We anticipate that this methodology may be of widespread utility to facilitate ligand optimisations in structure-based drug design efforts.

Conflicts of interest

There are no conflicts to declare.

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Notes and references


