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Using mass reconstruction along a four-site transect as a method to interpret PM$_{10}$ in west-central Scotland, UK

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ABSTRACT

Concurrent 24 hr samples of PM$_{10}$ were collected over a 10 day period in August 2000 at four sites along a transect in west-central Scotland, UK (passing from the coast through the city of Glasgow) in line with the prevailing south-westerly wind. Each sample was analyzed for, chloride (Cl$^-$), nitrate (NO$_3^-$), sulfate (SO$_4^{2-}$), ammonium (NH$_4^+$), calcium (Ca$^{2+}$), iron (Fe) and organic hydrocarbon material (OHM). The contribution from elemental carbon (EC) was estimated. Sampling days were categorized according to local wind direction, synoptic flow and air mass back trajectories. Chemical mass balance reconstruction of the following PM$_{10}$ components was derived for each wind direction group and at each transect location: ammonium sulfate ((NH$_4$)$_2$SO$_4$), ammonium nitrate (NH$_4$NO$_3$), sodium chloride (NaCl), gypsum (CaSO$_4$), OHM, EC, soil/surface dusts and particle bound water. The results showed that PM$_{10}$ at the coastal site was dominated by the marine background (NaCl) compared with the urban sites which were dominated by local primary (elemental carbon and soil/resuspension) and secondary sources (NH$_4$NO$_3$, (NH$_4$)$_2$SO$_4$ and OHM). There was evidence of chloride depletion as NaCl aerosol passes over urban areas. There was also evidence of long-range transport of primary PM$_{10}$ (EC and OHM), for example at the coastal site from transport from Ireland. The work demonstrates how the general approach of combining mass reconstruction along a transect with other information such as wind/air-mass direction generates insight into the sources contributing to PM$_{10}$ over a spatial scale more extended than at a single receptor.

IMPLICATIONS
This study highlights the utility of a multi-site transect aligned with the prevailing wind to identify changes in PM$_{10}$ chemical composition in air masses moving inland from a coastal region, through urban and peri-urban regions. Local, marine and long-range PM$_{10}$ components were identified. The chemical mass balance (CMB) calculation offers a pragmatic method for accounting for the major chemical sources contributing to PM$_{10}$ at a receptor. Interpreting the CMB of PM$_{10}$ is of policy relevance not only in helping identify sources but also because of the implications for health impacts.

**INTRODUCTION**

Numerous studies published over the last 20 years have demonstrated a significant, linear association between the concentration of airborne particulate matter (PM) of median aerodynamic diameter less than 10 µm (PM$_{10}$) and cardiopulmonary mortality and morbidity.\(^1\)\(^-\)\(^3\) There appears to be no lower threshold in the dose-response between PM$_{10}$ concentrations and health effects.\(^4\) Evidence suggests that particle size, number and chemical composition all ultimately determine the PM toxicological impact.\(^5\)

PM$_{10}$ has many sources including anthropogenic, biogenic, geogenic, primary and secondary, local and long-range.\(^6\)\(^,\)\(^7\) In order to determine the source contribution to PM$_{10}$ present at any given receptor it is necessary to gain as much knowledge as possible regarding the chemical composition, meteorological factors and an appreciation of the local and long-range sources impacting the receptor.\(^8\)\(^,\)\(^9\) A number of methods to apportion PM$_{10}$ sources have been applied. These include deterministic approaches utilizing emission inventories and forward process modeling of the subsequent expected atmospheric concentration.\(^10\) However these models are computationally and input-data demanding and often exclude contributions from...
resuspended mineral dusts and marine-derived salt particles. Other approaches use receptor models based upon multivariate statistics, including principal component analysis and the more sophisticated positive matrix factorization method.

The chemical mass balance (CMB) approach to receptor modeling offers a relatively simple method for source apportionment of airborne PM. Knowledge of the CMB of PM not only helps to elucidate their source but can provide insight into their potential toxicology effects. Additionally, the CMB information can aid air quality managers in the identification of particular anthropogenic emissions and subsequent management strategies to reduce them. A number of studies have used CMB to account for the collected PM mass through measured and inferred major chemical constituents. For example, a “pragmatic” mass closure model was applied to the source apportionment of PM \(_{1.0}\), PM \(_{2.5}\), and PM \(_{10}\) collected at roadside, urban background and rural sites within and outside of Birmingham, UK. This study concluded that the pragmatic CMB approach developed previously was suitable for applying to new data sets at new locations and times.

**AIM AND OBJECTIVES**

The aim of this study was to investigate chemical mass reconstruction as a method to gain insight into the factors (local emissions, long range transport, chemistry, etc.) influencing changes to PM \(_{10}\) chemical composition along a transect through a major conurbation in the UK, specifically a SW to NE transect through the towns of Prestwick, Paisley, Glasgow and Cumbernauld in west-central Scotland. This is the first analysis of PM \(_{10}\) for a transect in Scotland; with only one such analysis previously reported for the UK in SW England.
The determination of ammonium, nitrate, sodium, calcium, chloride, sulfate and iron extracted from collected PM$_{10}$ is relatively straightforward, with recognized standard methods well documented in the literature. $^{22-24}$ However, the measurement of organic carbon (OC) and elemental carbon (EC) is problematic and there is no internationally-agreed standard analytical procedure. $^{25}$ Some current methods use a combination of thermal and optical techniques to differentiate between the OC and EC fractions within a sample of PM, $^{26,27}$ but there are recognized problems with these methods. $^{25,28}$ An alternative approach to estimate EC is via filter reflectance, but this is not a well quantified metric and suffers from issues of colored metal oxides and overloaded filters. $^{29}$ In this work, a loss on ignition approach was used to quantify total organic hydrocarbon material without requirement for further scale-up of an OC mass to total organic material mass.

**METHODS**

**Sampling**

Site selection was based on a transect aligned with the prevailing south-westerly wind direction; $^{30}$ with sites of contrasting coastal and urban environments including sub-urban zones upwind and downwind of a major urban centre. The sampling locations are shown in Figure 1 and descriptive data in Table 1. Prestwick is a small coastal town with low vehicle density. The sampler was located ~240 m from the shore. Although Prestwick International Airport is 1.5 km to the NNE, this is predominantly downwind of the site, and the airport has low passenger numbers compared with Glasgow International Airport (PIA: 905,000 in 2000; GIA: 6,924,000). $^{31}$ Apart from the airport and local traffic there are few local anthropogenic sources of PM$_{10}$ in Prestwick. Paisley is a larger urban area; the PM$_{10}$ sampler was centrally located on the Town Hall veranda, 4 m above the ground, and surrounded by urban roads.
carrying a total of 10,000 vehicles day\(^{-1}\). The city of Glasgow is one of the largest urban conglomerations in the UK. The monitoring site was located in the centre of the city, but 30 m above street level in order to represent urban background. Glasgow has one of the highest vehicle densities in the UK due to the convergence of four major transport routes (M8, M80, M77 and M74). The M8 (which has the highest vehicle density) bisects Glasgow, the nearest point of which lies 860 m to the North West of the sampling site. There are two major railway termini in Glasgow which operate many diesel trains and are a considerable local source of elemental carbon. The predominantly downwind, sub-urban Cumbernauld monitoring site was located in a residential housing estate, adjacent to a road with a vehicle density of approximately 3,000 vehicles day\(^{-1}\).

Samples of PM\(_{10}\) were collected concurrently at the four sites using US EPA Federal Reference Method filter samplers (Thermo Partisol 2000) each fitted with a US EPA approved SA246b Graseby Andersen PM\(_{10}\) size-selective inlet and sampling at a flow rate of 16.7 l min\(^{-1}\). Samples were collected daily (starting at 8 am each day) from 7\(^{th}\) - 17\(^{th}\) August 2000.

**Gravimetric and chemical analysis**

Samples were collected onto 47 mm diameter Whatman QMA quartz fibre filter with 0.6 µm diameter pore size. All filters were baked for 4 hours at 900°C in a muffle furnace before use, then washed with 18 MΩ water, air dried in Petri dishes and placed in a desiccator to equilibrate for 48 h at 35-45% RH and 20 ± 2 °C prior to weighing. The filters, including blanks and six control filters, were weighed using a Sartorius MC5 microbalance following the protocol used by Heal et al. (2000). An antistatic dissipater was employed to eliminate
weighing artifacts due to electrostatic charge. Gravimetric imprecision was calculated to correspond to 1 µg m⁻³ for a 24 hr sample.

After re-weighing, the loaded filters were cut into thirds for chemical analysis: one third for the analysis of cations (Na⁺, Ca²⁺ and NH₄⁺) and anions (Cl⁻, NO₃⁻ and SO₄²⁻), one third for the analysis of Fe (and other trace metals), and one third for indirect quantification of organic hydrocarbon material via mass loss on ignition at 550°C. Blank filters were used to check for background contamination for all analyses.

The filter portions for anion and cation determination were extracted with sonication into 5 ml of 18 MΩ water and the solution analysed using a Dionex 2000i ion chromatography system with conductivity detection. For cation analysis, a Hamilton PRP-X200 separation column (Phenomenex) was used, with a mobile phase of 4 mM nitric acid in 3:7 methanol:water at 2 ml min⁻¹ flow rate. For anion analysis, a Dionex AS4A separation column preceded by a Dionex AG4A guard column was used, with a mobile phase of 1.8 mM Na₂CO₃ and 1.7 mM NaHCO₃ at 2.5 ml min⁻¹ flow rate. A Dionex AMMS-II Anion Micro Membrane Supresser was also employed, with a suppressor regenerant consisting of 25 mM H₂SO₄ at a flow rate of 6 ml min⁻¹. In both cases, sample injection volume was 100 µl. Iron and other trace metals were extracted from the filter using a mixed acid solution (5 ml) prepared by mixing concentrated nitric acid (65 ml), concentrated hydrochloric acid (185 ml) and diluting to 1 l with 18 MΩ water. Four drops of concentrated perchloric acid were added to each 5 ml extract solution to aid the extraction process. The extracts were heated to 40°C for 2 hr, followed by 1 hr at 90°C and 2 hr at 140°C, then diluted to 10 ml with 18 MΩ water and centrifuged at 10,000 rpm for 10 min. Metal concentrations were determined using ICP-OES (Jobin-Yvon 138) against 7-point standard calibration curves. The loss on ignition at 550°C
provides a good estimate of the organic hydrocarbon content of PM, \(^{25, 27}\) once corrected for loss of other volatile material (see later). It is recognized that some hydrocarbons may char rather than be driven off at this temperature and thus lead to an underestimation of OHM and an overestimation of EC (as derived here from the remaining “unaccounted for” mass of PM\(_{10}\)).

**Meteorological Data**

Temperature and relative humidity were recorded by each Partisol unit at each monitoring location. Other meteorological data (local wind direction, wind speed, and rainfall) were obtained from a Meteorological Station, 12.2 km west of the Glasgow monitoring site, 5.9 m above mean sea level. Synoptic charts were obtained from the Royal Meteorological Society (www.rmets.org) and used to classify wind direction according to “Lamb Weather Type” which contains eight main directional types: north, north-east, east, south-east, south, south-west, west and north-west.\(^{30}\)

In addition, 5-day air mass back trajectories were calculated for arrival every 6 hours at the 950 hPa pressure level at each site using software and data at the British Atmospheric Data Centre (badc.nerc.ac.uk). Air mass back trajectories are a useful tool for determining potential upwind sources of air pollutants reaching a receptor.\(^{33, 34}\) Trajectories were visualized using the AGIS Geographical Information System\(^{35}\) and categorized according to the broad geographical sector over which they had passed prior to reaching each site, with categorization defined by having at least 70% of the trajectory in that sector.\(^{30}\)

**RESULTS**
Meteorology

Generally there was little rainfall during the study period. The 7, 8, 11, 12, 14 & 17 August 2000 had little rain, ranging between 2 – 4 mm; the remaining days were dry. The relative humidity was unusually high, between 95 and 100% at all of the monitoring locations, compared with the normal range 71 to 90% for this location. The winds were predominately from the W and SW with fairly uniform wind speed around 3 m s\(^{-1}\).

Data groupings

Daily air mass back trajectories could be consigned as described above to one of the following four categories, illustrated in Figure 2.

- **WESTERLY**– 5 days spent over the Atlantic before making landfall in the vicinity of Prestwick.
- **SOUTH WESTERLY**– from the Atlantic, then crossing Ireland from SW to NE before crossing the Irish Sea before making landfall in the vicinity of Prestwick.
- **SOUTHERLY**– from the Atlantic, through the Irish Sea avoiding Ireland, England and Wales before making landfall in the vicinity of Prestwick.
- **EASTERLY**– originating in Eastern Europe crossing the North Sea and central Scotland prior to reaching the transect sites.

Scrutiny of the local wind direction plots, Lamb weather types and chemical compositional data revealed similarities for certain sampling events, e.g. for the 7 and 8 August. These days were characterized by a westerly wind and low levels of PM\(_{10}\) at Prestwick and Paisley. The air mass back trajectories confirmed similar geographical categorization. On this basis these
two days were grouped within the ‘Westerly’ classification. Days were grouped together only if both the meteorological parameters and the PM chemical composition were similar. This approach to grouping yielded the classification shown in Table 2. In the case of 14 August the local wind direction was highly variable and it was not possible to assign it into a well-defined category.

**Transect PM$_{10}$ gravimetric concentrations**

The descriptive statistics of the PM$_{10}$ concentrations at each location along the transect are illustrated as box plots in Figure 3.

**Mass Reconstruction**

The numerical procedures used to reconstruct PM$_{10}$ chemical composition from measured components are detailed in Table 3 and were based on the work of Harrison et al. (2003), but with minor modifications. First, ammonium content was measured explicitly in this work so it was possible to calculate contributions from each of (NH$_4$)$_2$SO$_4$, NH$_4$NO$_3$ and NaNO$_3$ independently. Secondly, total organic hydrocarbon material (OHM) was derived indirectly. The loss on ignition method drives off hydrocarbon species, but also drives off particle-bound water and volatilizes NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$ (which all have volatilization temperatures well below the 550°C temperature used). The mass of OHM in each sample is therefore the mass lost during the heating after subtraction of the masses of these latter components. Any other unidentified component that might also be lost on combustion will have small mass contribution compared to the major chemical species accounted for in Table 3. After accounting for all the other components detailed in Table 3, the remaining mass in each PM$_{10}$
sample was assigned as EC. Thus, the CMB approach used here reconstructs the entire PM\textsubscript{10} mass by definition. As discussed in the introduction, any organic hydrocarbon material that chars in situ on the filter rather than being driven off will erroneously be assigned as EC, but this is an issue also for more sophisticated thermo-optical OC-EC analysers.

The PM\textsubscript{10} reconstructed mass (as defined above), for each site, and for each wind direction categorization, are shown in Figure 4, and expressed as percentage fractions of each reconstructed mass in Figure 5.

**PM\textsubscript{10} Dispersion along the Transect**

SCREEN VIEW, a US EPA Gaussian Dispersion Screening model,\textsuperscript{37} was used to assess the dispersion of primary surface emissions of PM\textsubscript{10} from an upwind site downwind to the next site along the transect. The meteorology observed during SW airflow was used in the model as this is in alignment with the transect sites. The model assumes that PM\textsubscript{10} does not change state over the timescale modelled and that wind direction remains constant. The model does not include background and/or long-range transported PM\textsubscript{10} sources. Additionally, only simple terrain can be incorporated into the model whereas there are hills between Prestwick and Paisley. The terrain from Paisley to Glasgow to Cumbernauld does however fit the ‘simple’ category. Despite these simplifications, the model provides some useful insight into the extent to which dispersion of primary PM\textsubscript{10} emissions at the upwind site can impact the downwind receptor. Figure 6 illustrates the modelled dispersion of primary emissions of PM\textsubscript{10} in Glasgow downwind towards the Cumbernauld site. The modelled contributions, on average, to the total PM\textsubscript{10} concentrations at each site due to primary PM\textsubscript{10} emissions from the
adjacent upwind urban area were as follows: Prestwick to Paisley = 0.34 µg m⁻³, Paisley to Glasgow = 3.6 µg m⁻³, Glasgow to Cumbernauld = 8.5 µg m⁻³.

DISCUSSION

Although the discussion here necessarily focuses on the data obtained in this study, it demonstrates the general approach that can be taken to analyzing PM compositional data from along a transect.

A multivariate analysis of variance confirmed the visual impressions gained from Figures 3-5; namely that sampling site has a significant effect upon the mean concentrations of NaCl, (NH₄)₂SO₄, NH₄NO₃, NaNO₃, OHM, soil and EC in sampled PM₁₀ (𝑝 = 0.015), and that wind direction also has a significant effect upon the mean concentrations of the same components (𝑝 = 0.014). The interpretations of the detail of the observed differences in PM₁₀ composition sampled along the transect are given below.

Figure 3 shows that the highest average PM₁₀ observed was in Cumbernauld (X ± 1 SD = 40.0 ± 10 µg m⁻³) and Glasgow (30 ± 15 µg m⁻³). As shown in Figure 4 and 5, PM₁₀ at these two locations are dominated by EC, OHM, Soil/Dust, NH₄NO₃, (NH₄)₂SO₄, and associated particle-bound water. The lowest PM₁₀ concentrations were observed at Prestwick under the conditions of the Westerly air flow prevailing on 7 and 8 August (7.6 µg m⁻³), the South Westerly air flow prevailing on 11, 15, 16 & 17 August (7.3 µg m⁻³) and the Southerly air flow prevailing on 12 August (8.8 µg m⁻³). Figures 4 and 5 show that under Westerly air flow the PM₁₀ in the coastal town of Prestwick is dominated by NaCl, presumably of marine origin from the Atlantic. The “outlier” PM₁₀ concentration for Paisley shown in Figure 3 is
terminology assigned by the plotting package and should not be interpreted as erroneous since
the gravimetric value is consistent with the independent measurements of the major chemical
components of this sample. This datum is for the 13 Aug, the one day of easterly air flow,
when air masses arriving at Paisley had previously passed over Glasgow, Central Scotland
and the European mainland.

The composition of PM$_{10}$ at Prestwick and Paisley, for EC, OHM, NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$,
was highly significant different ($p < 0.0005$) under Easterly air flow compared to all the other
wind direction groups (Figures 4 and 5). This suggests upwind sources of primary and
secondary ‘aged’ aerosol originated from continental Europe and Central Scotland. Figure 4
and 5 also show evidence of transport of EC and OHM advected from Ireland during the
period 11, 15, 16 and 17 August; these components have low concentrations at Prestwick for
westerly air flow directly from the North Atlantic but missing Ireland (7 and 8 August), but
show a highly significant increase ($p < 0.0005$) during south-westerly air flow which has
crossed Ireland and passed over the city of Belfast en route to Prestwick. Westerly air flow on
10 August appears also to have entrained some secondary aerosol (as revealed by NH$_4$NO$_3$)
before arrival at Prestwick. The air mass back trajectories showed that the air mass arriving on
the 10 August passed close to the North of Ireland coastline as compared with the Westerly
airflows on the 7 and 8 August. Despite this, PM$_{10}$ on 10 August at Prestwick and Paisley still
contained predominant proportion from NaCl, demonstrating the significant contribution that
marine aerosol makes to the total PM$_{10}$ observed in maritime regions. Figures 4 and 5 also
show that EC, OHM, NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$ concentrations at Prestwick and Paisley are
somewhat elevated during Southerly air flow (e.g. 9 August) compared with Westerly and
South-Westerly air flow, presumably due to advection of PM$_{10}$ from England to all the sites
under these conditions.
An interesting observation is that, for most of the sampling campaign, the NaCl content of the PM$_{10}$ declines substantially between Paisley and Glasgow. It is assumed this is due to the numerous reactions that are known to occur between NO$_2$, H$_2$SO$_4$, HNO$_3$, NH$_4$NO$_3$, (NH$_4$)$_2$SO$_4$ and NaCl which lead to removal of Cl from the particle phase via formation of gaseous HCl and ClNO$_2$.\textsuperscript{38,39} for example, NaCl$_{(s)}$ + HNO$_3(g)$ $\rightarrow$ NaNO$_3(s)$ + HCl$_{(g)}$. The loss of NaCl at Glasgow and Cumbernauld is supported by the greater NH$_4$NO$_3$ and (NH$_4$)$_2$SO$_4$ concentrations in Glasgow and Cumbernauld than Prestwick and Paisley (except for Easterly air flow where they are comparable due to advection from more polluted terrestrial areas).

Another interesting observation is the general trend for increased “soil/resuspended” component concentration from Prestwick $\rightarrow$ Paisley $\rightarrow$ Glasgow (Figure 4); this source contributes $\sim$20% of the PM$_{10}$ composition in Glasgow and Cumbernauld. The increasing trend is evidence that the soil component originates principally from local sources during passage along the transect, most likely vehicle-induced re-suspension of surface soil on roads from the increased traffic density in the Glasgow and Cumbernauld area. There was no discernable site specific or wind directional trend associated with the gypsum component, since this is a marker of site-specific construction activity.

A further feature is the increase in the OHM/EC ratio moving inland from the coast. The OHM/EC ratios for all Prestwick and Paisley samples ranged from 0.20 to 0.88, whereas the OHM/EC ratios for Glasgow and Cumbernauld ranged from 0.89 to 1.00. The OHM/EC ratios observed for Glasgow and Cumbernauld are similar to those reported for other urban environments.\textsuperscript{25,40} Figure 5 shows that the OHM has some spatial uniformity in Prestwick, Glasgow and Cumbernauld, accounting for approximately 20%, on average, of total PM$_{10}$.
This is also consistent with the findings of Jones and Harrison (2005) who observed that OC contributes ~22%, on average to total PM$_{10}$ with reasonable spatial uniformity across other sites in the UK. The OHM will comprise both locally primary emitted OHM and secondary OHM generated on the regional scale. The lightly traffic influenced coastal Prestwick site in this study shows lower OHM concentrations.

The simple Gaussian dispersion modelling showed that approximately 8 - 9 µg m$^{-3}$ of primary-emitted PM$_{10}$ may be advected to Cumbernauld from Glasgow (for Westerly and South Westerly wind directions). Although Figure 6 shows that primary PM$_{10}$ emissions disperse rapidly in the first 10 km downwind, Glasgow is within this 10 km zone from Paisley so does inherit a significant PM$_{10}$ contribution from Paisley (~12% of the mean PM$_{10}$ concentration in Glasgow). Similarly, Cumbernauld, although 17 km downwind of Glasgow, receives considerable primary PM$_{10}$ import from Glasgow approximately 20% of the total on average, due to the high primary emissions within Glasgow. These observations obviously have implications for effectiveness of local air quality management strategies.

Although the data for this study were collected a number of years previously and emissions of primary PM, and of precursors contributing to secondary PM, have generally continued to decline in the UK and most of Europe, the major sources and atmospheric processes contributing to receptor-measured PM$_{10}$ in the UK are broadly unchanged. Thus whilst the primary purpose here is to demonstrate a methodological approach, it is anticipated that the specific results for the example study area remain valid also.
CONCLUSIONS

This work demonstrates how simultaneous sampling of PM$_{10}$ along a transect, coupled with chemical mass balance, provides insights into the changes in PM$_{10}$ chemical composition with prevailing wind direction; in this study, for example, on moving inland from the west coast of Scotland through the major conurbation of Glasgow. The upwind coastal site at Prestwick revealed the marine and long-range transport components whilst PM$_{10}$ in the city of Glasgow, and at the downwind site near Cumbernauld included substantial primary material associated with high-density urban areas (EC, OHM, resuspended material). There was clear evidence of chloride loss from PM$_{10}$ moving inland. Under Easterly air flow, a component of long-range transport of secondary aged aerosol (NH$_4$NO$_3$, (NH$_4$)$_2$SO$_4$) from continental Europe was apparent. Studies of the design reported can contribute to policy development by helping identify sources contributing to PM$_{10}$ over a spatial scale more extended than at a single receptor and thus to more informed targeting of abatement measures.

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Table 1. Summary of PM$_{10}$ monitoring site characteristics

<table>
<thead>
<tr>
<th>Location</th>
<th>Site Characteristics</th>
<th>Site coordinates</th>
<th>Distance from adjacent site running SW to NE (km)</th>
<th>Elevation (m)</th>
<th>Pop$^a$</th>
<th>Area (km$^2$)</th>
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</thead>
<tbody>
<tr>
<td>Prestwick</td>
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<td>0</td>
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<td>30.0</td>
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<td>Cumbernauld</td>
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<td>17</td>
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Table 2. Data group categories

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<tr>
<td>Westerly air flow</td>
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<tr>
<td>Westerly air flow</td>
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<tr>
<td>South Westerly air flow</td>
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<tr>
<td>Southerly air flow</td>
<td>9</td>
</tr>
<tr>
<td>Southerly air flow</td>
<td>12 (spanning Saturday &amp; Sunday)</td>
</tr>
<tr>
<td>Easterly air flow</td>
<td>13</td>
</tr>
</tbody>
</table>

14 August could not be categorized due to too much deviation in the air flow.
Table 3. Adjustment factors used in mass closure (adapted from Harrison et al., 2003)

<table>
<thead>
<tr>
<th>Analyte</th>
<th>Conversion to</th>
<th>Numerical factor</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sulfate</td>
<td>(NH$_4$)$_2$SO$_4$</td>
<td>Take measured mass of SO$_4$ and subtract (2.4x Ca) to yield a corrected SO$_4$ not contained within Gypsum. Multiply result by 1.375</td>
</tr>
<tr>
<td>Particle Bound Water</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ammonia</td>
<td>NH$_4$NO$_3$</td>
<td>Take measured mass of NH$_4$ and subtract (0.375x corrected mass of SO$_4$ as above) to yield NH$_4$ not associated with (NH$_4$)$_2$SO$_4$. Multiply result by 4.44 to yield NH$_4$NO$_3$</td>
</tr>
<tr>
<td>Particle Bound Water</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Nitrate</td>
<td>NaNO$_3$</td>
<td>Take measured mass of NO$_3$ and subtract (0.775x derived mass of NH$_4$NO$_3$) to yield NO$_3$ not associated with NH$_4$NO$_3$. Multiply result by 1.375 to yield NaNO$_3$</td>
</tr>
<tr>
<td>Chloride</td>
<td>NaCl</td>
<td>Multiply Cl by 1.648</td>
</tr>
<tr>
<td>Calcium</td>
<td>CaSO$_4$.2H$_2$O (Gypsum)</td>
<td>Multiply Ca by 4.30</td>
</tr>
<tr>
<td>Iron</td>
<td>Soil/road dust</td>
<td>Multiply Fe by 7.50</td>
</tr>
<tr>
<td>Mass lost on filter</td>
<td>Organic Hydrocarbon Material</td>
<td>Take mass of material lost on combustion and subtract (mass of (NH$_4$)$_2$SO$_4$ + NH$_4$NO$_3$ + associated particle bound water).</td>
</tr>
<tr>
<td>Unexplained mass</td>
<td>Elemental Carbon</td>
<td>Remaining mass ~ elemental carbon</td>
</tr>
</tbody>
</table>
Figure 1: Map of the PM10 sampling locations.
Figure 2: Four wind direction groups defined during sampling between August 7 and 17, 2000.
Figure 3: Box plots of the PM$_{10}$ concentrations measured at each site in the transect over the period August 7–17, 2000.
Figure 4: PM$_{10}$ reconstructed mass by site and by wind direction group (August 7–17, 2000).
Figure 5: Percent reconstructed PM10 mass by site and wind direction group (August 7–17, 2000).
Figure 6: Screen view modeled PM$_{10}$ dispersion downwind of Glasgow during southwesterly airflow. The initial rise in PM$_{10}$ concentrations arises because Glasgow is a large area source and air continues to pick up PM$_{10}$ primary emissions downwind from the center of the city before dispersion dominates beyond the conurbation boundary.