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Inelastic neutron scattering and frequency-domain magnetic resonance studies of \( S=4 \) and \( S=12 \) \( \text{Mn}_6 \) single-molecule magnets

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We investigate the magnetic properties of three \( \text{Mn}_6 \) single-molecule magnets by means of inelastic neutron scattering and frequency domain magnetic resonance spectroscopy. The experimental data reveal that small structural distortions of the molecular geometry produce a significant effect on the energy-level diagram and therefore on the magnetic properties of the molecule. We show that the giant spin model completely fails to describe the spin-level structure of the ground spin multiplets. We analyze theoretically the spin Hamiltonian for the low-spin \( \text{Mn}_6 \) molecule \((S=4)\) and we show that the excited \( S \) multiplets play a key role in determining the effective energy barrier for the magnetization reversal, in analogy to what was previously found for the two high spin \( \text{Mn}_6 \) \((S=12)\) molecules [S. Carretta et al., Phys. Rev. Lett. 100, 157203 (2008)].

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I. INTRODUCTION

Single-molecule magnets (SMMs) have been the subject of intense research activity since the first and mostly studied one, \( \text{Mn}_{12}\text{Ac} \) was reported.1 These metal-organic clusters are usually characterized by a large spin ground state \( S \) and an easy-axis anisotropy which determines the zero-field splitting (ZFS) of the \( S \) state sublevels. The resulting magnetic bistability makes them interesting for magnetic storage applications due to their potential to shrink the magnetic bit down to the size of one single molecule. Until recently and despite the common efforts of chemists and physicists to find suitable systems that could retain the magnetization for a long time at noncryogenic temperatures, \( \text{Mn}_{12}\text{Ac} \) was the system showing the “highest” blocking temperature (3.5 K) and anisotropy barrier (74.4 K).2 The relaxation time in the classical regime follows the Arrhenius law: \( \tau = \tau_0 \exp(U/k_B T) \) (Ref. 3). According to this, there are two key points that have to be considered for the realization of an ideal SMM. First of all, the anisotropy barrier, given to a first approximation by \( U \sim |D|S^2 \) (\( D \) is the axial anisotropy parameter), has to be sufficiently high. This is to prevent the reversal of the magnetization via a classical thermally activated multistep Orbach process mediated by spin-phonon interactions. This can be achieved by the simultaneous increase in \( D \) and \( S \), two variables that are intrinsically linked together.4 Second, the pre-exponential factor \( \tau_0 \) in the Arrhenius law has to be large. This factor is dominated by the time necessary to climb the upper states in the energy-level diagram and is proportional to \( D^{-3} \) (Refs. 3, 5, and 6).

In addition to the classical relaxation mechanism, the quantum tunneling of the magnetization that characterizes the spin dynamics of SMMs, has to be taken into consideration and minimized for magnetic data storage application, since it provides a shortcut for the relaxation of the magnetization.

Therefore, to engineer SMMs able to retain the magnetization for long time it is crucial to control all the different mechanisms that provide a relaxation path for the system. Recently we succeeded in the synthesis of a new class of \( \text{Mn}^{3+} \)-based clusters that contributed in raising the anisotropy barrier and has served as a good model system to study the factors involved in the relaxation mechanism.7,8

This class consists of hexanuclear \( \text{Mn}^{3+} \) clusters (from now on \( \text{Mn}_6 \)) which, despite the generally similar nuclear structure, display a rich variety of spin ground states and anisotropy energy barriers.9–12 The six \( \text{Mn}^{3+} \) ions are arranged in two triangles, with dominant ferromagnetic (FM) exchange interaction between the two triangles and FM or antiferromagnetic (AFM) interactions within the two triangles. It has been found that the nature of the intratriangle exchange interaction can be switched from AFM to FM by substituting the organic ligands bridging the \( \text{Mn}^{3+} \) ions, leading to a change in the ground state from a low spin \((S=4)\) to a high spin \((S=12)\).9 Furthermore, deliberately targeted structural distortions have been successfully used to tune the values of the exchange interactions.10 The isotropic exchange interactions, and consequently the overall anisotropy barrier, is thus found to be very sensitive to the structural details. This has been also demonstrated using an alternative method for distorting the molecule, that is, by applying external hy
changes from antiferromagnetic in ferromagnetic for all molecules, the intratriangular coupling. For example, while the coupling between the two triangles is FM. For this reason, even small structural distortions have sensitive to intramolecular bond angles and distances. The par-oxygen and nitrogen ions and is found to be extremely sensive for the desired magnetic properties.

We have investigated three members of the family of Mn₆ clusters, with chemical formulas (1) [Mn₆O₂(sao)₁₀ (O₂CMe)₂2(EtOH)₄].4EtOH, (2) [Mn₆O₂(Et-sao)₁₀(O₂CPh)₂ (EtOH)₁₁(H₂O)₁₂].2EtOH, and (3) [Mn₆O₂(Et-sao)₁₀ [O₂CPh(Me)₂]₂(EtOH)₁₁].9,10,16 All molecules display very similar structures consisting of six Mn³⁺ ions (s = 2) arranged in two staggered triangular units (see Fig. 1) related by an inversion center.

The only major structural difference between the three clusters resides is the steric effect of the organic ligands used in proximity to the transition-metal ions. However, despite having very similar structures, the three molecules have very different magnetic properties. The coupling between the magnetic ions occurs via superexchange pathways involving oxygen and nitrogen ions and is found to be extremely sensitive to intramolecular bond angles and distances. The par-ticular arrangement of the magnetic ions provides exchange couplings lying in the crossover region between AFM and FM. For this reason, even small structural distortions have tremendous impact on the magnetic properties of the system. For example, while the coupling between the two triangles is ferromagnetic for all molecules, the intratriangular coupling changes from antiferromagnetic in (1) to ferromagnetic in (2) and (3) due to a “twisting” of the oximate linkage. This results in a “switching” of the total spin ground state from S = 4 to S = 12. Systematic synthesis and studies of various members of the Mn₆ family have revealed that the nature of the coupling is extremely sensitive to the intratriangular Mn-O-N-Mn torsion angles12,13 (see Fig. 1). There is a critical value for the torsion angle of 30.85° ± 0.45°, above which the pairwise exchange interaction switches from antiferromagnetic to ferromagnetic, while a further enhancement of the angle increases the strength of the FM interaction. This effect has been interpreted in terms of the particular arrange-ment of the manganese d₃ orbitals with respect to the p orbitals of the nitrogen and oxygen ions. A large (small) Mn-O-N-Mn torsion angle results in a small (large) overlap between the magnetic orbitals giving rise to ferromagnetic (antiferromagnetic or weak ferromagnetic) superexchange interactions.17

Molecules (2) and (3) have the same spin ground state S = 12 but very different effective energy barriers Ueff ≈ 53 K for (2) and Ueff ≈ 86.4 K for (3). This differ-ence was found to be closely related to the exchange interactions.7

In order to understand this rich variety of behaviors, we performed a detailed spectroscopic characterization of the three molecules using inelastic neutron scattering (INS) and frequency domain magnetic resonance (FDMR). FDMR is only sensitive to transitions with a predominantly intramulti-plet characteristic, according to the selection rules ΔS = 0, ΔMₛ = ± 1. In contrast, in INS both intermultiplet and intramultiplet transitions can be observed (ΔS = 0, ± 1, ΔMₛ = 0, ± 1). Thus, the combination of the two techniques allows assignment of all observed exci-tations.18,19

The determination of the model spin Hamiltonian parameters enabled us to estimate theoretically the effective energy barrier for the low spin molecule (1). Similarly to what we previously reported for the two high spin molecules (2) and (3), the results on (1) show how the presence of low-lying excited spin multiplets plays a crucial role in determining the relaxation of the magnetization.

In conventional systems, the effects of S mixing can be effectively modeled by the inclusion of fourth order zero-field splitting parameters in the giant spin Hamiltonian.20 Here we will show that this Hamiltonian is completely inad-equate for the description of the spin-state energy-level structure.

II. EXPERIMENTAL METHODS

Nondeuterated polycrystalline samples were synthesized according to published methods9,10 FDMR spectra were re-coded on a previously described quasioptical spectrometer,21 which employs backward wave oscillators as monochromatic coherent radiation sources and a Golay cell as detector. Sample (1) proved to deteriorate rapidly upon pressing and over time. Therefore, the FDMR measurements on (1) were performed on loose microcrystalline material (348 mg) held between two quartz plates. In this unconven-tional measurement, the detector signal was recorded as function of frequency at different temperatures. Extreme care had to be taken to prevent the slightest positional changes in sample and equipment, which changes the standing wave pattern in the beam, precluding normalization. The normalized transmission was calculated by dividing the signal intensity at a given temperature by that at the highest temperature (70 K). Samples (2) and (3) deteriorate to a lesser extent and FDMR spectra were recorded on pressed powder pellets made by pressing ca. 250 mg of the unground sample with approximately 50 mg n-eicosane (to improve pellet quality) into a pellet. All spectra were simulated using previously described software.22
INS experiments were performed using the multidisk-chopper time-of-flight spectrometers V3/NEAT at the Helmholtz-Zentrum Berlin für Materialien und Energie (HZB, Berlin, Germany) and IN5 and IN6 at the Institute Laue Langevin (Grenoble, France). The samples were inserted into hollow cylindric-shaped Aluminum containers and mounted inside a standard orange cryostat to achieve a base temperature of 2 K. A vanadium standard was used for the detector normalization and empty can measurements were used for the background subtraction.

III. THEORETICAL MODELING AND EXPERIMENTAL RESULTS

The experimental data have been modeled using both the giant spin Hamiltonian (GSH), which considers the ZFS of the ground-state multiplet only, and the microscopic spin Hamiltonian, which treats isotropic exchange and single-ion ZFS at the same level. Including only ZFS terms, the giant spin Hamiltonian for a spin state $S$ reads

$$H_S = D_S S_z^2 + E_S (S_x^2 - S_y^2) + B^0_i S_i^0,$$

where $D_S$ and $E_S$ are second-order axial and transverse anisotropy, respectively, and $B^0_i$ is the fourth-order axial anisotropy, with $S_i^0$ the corresponding Stevens operator. The microscopic spin Hamiltonian includes an isotropic exchange term for each pairwise interaction and single ion ZFS terms for each ion

$$H = \sum_{i<j} J_{ij} s(i) \cdot s(j) + \sum_i d_i S_i^2 + \sum_i \{35 c_i z_i^2(i) + c_i [25 - 30 \sigma(s+1)] z_i^2(i)\},$$

where $s(i)$ are spin operators of the $i$th Mn ion. The first term is the isotropic exchange interaction while the second and third terms are the second- and fourth-order axial single-ion zero-field splitting, respectively (the $z$ axis is assumed perpendicular to the plane of the triangle). The spin Hamiltonians have been numerically diagonalized by exploiting the conservation of the $z$ component of the molecular total spin and the exchange anisotropy parameters have been varied to obtain a best fit of the experimental data.

A. Mn$_6$ (1) ($S$=4) $U_{\text{eff}}$=28 K

Sample (1) was the first reported member of the Mn$_6$ family. The building block of the molecule is the [Mn$_3$O$_3$] triangular unit where Mn$_2$ pairs, bridged by the NO oxime, form a Mn-O-N-Mn moiety (see Fig. 2).

The Mn-O-N-Mn torsion angles within each triangle are 10.7°, 16.48°, and 22.8°, giving rise to a dominant antiferromagnetic exchange coupling. The two triangular units are coupled ferromagnetically, resulting in a total spin ground state of $S$=4. Four out of the six metal ions (Mn1, Mn2, Mn1’, and Mn2’) are six-coordinate and in distorted octahedral geometry (MnO$_2$N), with the Jahn-Teller axis almost perpendicular to the plane of the triangle, while the two remaining ions (Mn3 and Mn3’) are five-coordinate and in square pyramidal geometry (see Fig. 2).

The higher-frequency resonance line is attributed to the transition from the $|S=4, M_s=\pm4\rangle$ to $|S=4, M_s=\pm3\rangle$ states. INS measurements have found to be necessary to unambiguously identify the origin of the lower-frequency transitions (see below). Assuming that these transitions are transitions within the ground multiplet, a fit of the giant spin Hamiltonian ZFS parameters [Eq. (1)] to the observed resonance line energies yields $D_{3=4} = -2.12 \pm 0.03$ cm$^{-1}$.
FIG. 3. (Color online) (a) FDMR spectra of unpressed polycrystalline powder of (1) recorded at various temperatures. The intensity of the higher-frequency resonance line decreases with temperature while that of the lower-frequency lines increases up to 30 K, beyond which it decreases again. Dotted lines indicate resonance lines due to impurities. (b) Expanded view of the low-frequency part of the 30 K spectrum. (c) Experimental and fitted spectrum at $T = 30$ K using the GSH approximation. Note the logarithmic scale in (a) and (c).

$(-0.263 \pm 0.004 \text{ meV})$ and $h^2 = +(1.5 \pm 0.5) \times 10^{-4}$ cm$^{-1}$ $(1.24 \pm 0.06 \times 10^{-3}$ meV). This ground state $S_3$ value is much larger than reported spectroscopically determined $D_3$ parameters for other manganese SMMs, e.g., $D_{\text{Mn12Ac}} = -0.457$ cm$^{-1}$ for Mn$_{12}$Ac, $D_{\text{Mn3Zn}} = -0.247$ cm$^{-1}$ for Mn$_{3}$Zn$_{2}$, and $D_{\text{Mn5}} = -1.16$ cm$^{-1}$ for Mn$_{5}$.

The main reason for this large $D$ value is the fact that the projection coefficients for the single ion ZFS onto the cluster ZFS are larger for spin states with lower $S$ (Ref. 27). The determined $D_{\text{Mn5}} = -2.12$ cm$^{-1}$ value for (1) is in excellent agreement with that found from density-functional-theory (DFT) calculations ($D = -2.15$ cm$^{-1}$). The expected energy barrier toward relaxation of the magnetization calculated from the found spin Hamiltonian parameters is $U_{\text{relax}} = |D|S^2 = 48.8$ K, which is much larger than the experimentally found $U_{\text{relax}} = 28$ K, indicating that more complex relaxation dynamics characterize this system, in analogy to what has been found for the Mn$_{5}$ S=12 compounds. The linewidth of the 1.33 meV line is slightly larger than that of the 1.80 meV line (48 $\mu$eV versus 41 $\mu$eV), which can indicate the presence of more than one excitation. The simulated spectrum agrees very well for the higher-frequency resonance line (note that the intensity is not rescaled) while the lower-frequency line is much weaker in the experiment than from the fit. This can be tentatively attributed to the presence of low-lying excited states as observed previously for Mn$_{12}$Ac. To determine the energy of excited spin states and identify the origin of the low-frequency resonances we resorted to INS, the technique of choice to directly access intermultiplet excitations.

The INS experiments were performed on $\approx 4$ g of non-deuterated polycrystalline powder of (1), which was synthesized as described in Ref. 16. For our measurements we used incident neutron wavelengths ranging from 3.0 to 5.92 Å with energy resolution between 50 and 360 $\mu$eV.

Figure 4(a) shows the INS spectra for an incident wavelength of $\lambda = 4.6$ Å (NEAT) for $T = 2$ K (blue circles) and $T = 20$ K (red squares). The continuous lines represent the spectra calculated assuming a dimer model for the spin Hamiltonian [Eq. (3)]. (b) $Q$ dependence of first intermultiplet (green circles) and intermultiplet (black squares) transitions measured on IN6 for $\lambda = 4.1$ Å and $T = 2$ K. Continuous lines represent the calculated $Q$ dependence using the dimer spin Hamiltonian, Eq. (3), (assuming a dimer distance of $R = 5.17$ Å, which corresponds to the distance between the center of the two triangles).
At $T=20$ K, we detected additional excitations at 1.05(1) and 1.31(1) meV, which must be due to transitions from excited states. All peaks in the INS spectra show a very unusual asymmetric line shape, which we assign to lattice solvent loss (see above).

From the comparison of INS data with the FDMR results, we can deduce that the excitation at 2.53 meV has a pure intermultiplet origin, being absent in the FDMR spectra (see Fig. 3). This is also confirmed by the $Q$ dependence of the scattering intensity of the observed excitations. Figure 4(b) shows this dependence for the $|S=4, M_s=\pm4\rangle \rightarrow |S=4, M_s=\pm3\rangle$ and $|S=4, M_s=\pm4\rangle \rightarrow |S=3, M_s=\pm3\rangle$ transitions. A characteristic oscillatory behavior has been observed for the $Q$ dependence of the intermultiplet INS transition (black squares), which presents a maximum of intensity at a finite $Q$ value (that is related to the geometry of the molecule), and decreasing intensity as $Q$ goes toward zero. This $Q$ dependence is typical for magnetic clusters and reflects the multispin nature of the spin states.30,31 By contrast, the intramultiplet excitation (circles) has maximum intensity at $Q=0$, as expected for a transition with $\Delta S=0$, and the intensity decreases with increasing $Q$, following the magnetic form factor.

The INS data directly reveal the presence of low-lying excited multiplets. Indeed, the difference in energy between the lowest and the highest energy levels of the anisotropy split $S=4$ ground state is given, as a first approximation, by $|D|S^2=4.2$ meV. The presence of an intermultiplet excitation at only 2.53 meV energy transfer, therefore below 4.2 meV, indicates that the first excited $S$ multiplet lies within the energy interval of the anisotropy split $S=4$ state. This suggests that the observed low-energy excitations are possibly not pure intermultiplet transitions but are expected to originate from the $S=4$ ground state and from the first excited $S$ multiplet. Therefore the exact assignment of those excitations requires a more accurate analysis beyond the GSH approximation. Indeed, one fundamental requirement for the validity of the GSH approximation, i.e., an isolated ground state well separated from the excited states, is not fulfilled and $S$ is not a good quantum number to describe the ground state of the molecule. To model the data it is thus necessary to use the full microscopic spin Hamiltonian of Eq. (2). This situation is also encountered in other studied molecules, such as, for example, in $\text{Mn}_2$32–36 in the $\text{Mn}_{[3 \times 3]}$ grid,37,38 in $\text{Ni}_2$,39 and in $\text{V}_{15}$,33,35,40 where the detailed modeling of the experimental results has required the use of a multispin approach.

Given the low symmetry of the triangular units in (1), the number of free parameters in Eq. (2) would be too large to obtain unambiguous results, considering the low number of experimentally observed excitations. Hence, we have chosen to describe the low-energy physics of (1) by a simplified dimer model, an approximation which has already previously been adopted for (3) (see Ref. 41). The dimer model was found to reproduce correctly the experimental results for molecule (3) and the calculated low-energy spectrum was found to be consistent with the one obtained using the microscopic spin Hamiltonian. In molecule (1) the small Mn-O-N-Mn torsion angles suggest a large dominant antiferromagnetic interaction within each triangle, as predicted by DFT calculations.17 The two triangular units can be therefore described as two ferromagnetically coupled $S=2$ spins, which also experience an effective uniaxial crystal-field (CF) potential.

The spin Hamiltonian has been diagonalized numerically and the $J$ and $d$ parameters have been varied to obtain a best fit of the experimental data. The position of the peak at 1.77 meV does not depend on the exchange interaction, therefore its position sets the value of the axial anisotropy $d$ parameter. Given the $d$ parameter, a fit of the position of the peak at 2.53 meV sets the isotropic exchange parameter $J$.

The best fit of the experimental data is obtained with $J=-0.19$ meV and $d=-0.59$ meV. The calculated energy-level scheme is reported in Fig. 5 (left), where the comparison with the energy-level diagram in the GSH approximation is also reported (right). The value of $S_{\text{eff}}$, where $S^2_{\text{eff}}=S_{\text{eff}}(S_{\text{eff}}+1)$, is labeled in color and shows that the first $S=3$ excited state is completely nested within the $S=4$ ground state. From Fig. 5 it is also clear that the GSH model does not account for a number of spin states different from the ground state $S=4$ multiplet at low energy. Furthermore, the assignment of the observed excitations can be misleading if considering the GSH approximation only. For example, using the GSH model, the observed peak at 1.33 meV can only be attributed to a pure intermultiplet excitation from $|4, \pm 3\rangle$ to $|4, \pm 2\rangle$, while using Eq. (3), it is found to be a superposition of several intermultiplet and intramultiplet transitions (indicated by arrows in Fig. 5). The GSH approximation fails to describe the low energy-level diagram of the molecule and consequently fails to describe the relaxation of the magnetization. Indeed, the presence of excited states nested within the ground state multiplet has a significant effect on the relaxation dynamics, as discussed in Sec. IV.
Introducing sterically more demanding oximate ligands results in a twisting of the Mn-N-O-Mn torsion angle, which causes switching of the intratriangle exchange interactions from antiferromagnetic to ferromagnetic, resulting in a large increase in the spin of the ground state from S=4 to S=12. Here, we study two [(2) and (3), respectively] of the many published derivatives of these S=12 Mn₆ clusters. Compound (2) has undergone two structural changes compared to (1). First of all, the distance between the phenolato oxygen and the two square pyramidal Mn₃⁺ ions has decreased from ≈3.5 to ≈2.5 Å, thus all Mn₃⁺ ions are now in six-coordinated distorted octahedral geometry (see Fig. 6). Second, the torsion angles of the Mn-N-O-Mn moieties has increased strongly with respect to those in (1), being 38.20°, 39.9°, and 31.26°, compared to 10.7°, 16.48°, and 22.8° for (1). In (3), the introduction of two methyl groups on the carboxylate ligand has increased the nonplanarity of the Mn-N-O-Mn moieties further, giving torsion angles of 39.08°, 43.07°, and 34.88°, respectively. The result is that the weakest ferromagnetic coupling is significantly stronger for (3) compared to (2). Using a single J model (e.g., assuming that the intratriangle and intertriangle exchange couplings are equal), Milios et al. fitted the dc susceptibility data for molecules (2) and (3) and obtained: J(2) = −0.230 meV and J(3) = −0.404 meV, respectively, (in our notation for the spin Hamiltonian).

In spite of the fact that both (2) and (3) have S=12 ground states and similar geometrical structures, radically different effective energy barriers toward the relaxation of the magnetization were observed, being U_{eff} ≈ 53 K for (2) and U_{eff} ≈ 86.4 K for (3). Here, we aim to understand this difference by an in-depth study of the energy-level structure by means of FDMR and INS.

Figure 7 shows FDMR spectra recorded on a pressed powder pellet of (2) at different temperatures. The baseline shows a pronounced oscillation, which is due to Fabry-Pérot type interference within the plane-parallel pellet. The oscillation period and downward slope to higher frequencies are determined by the thickness of the pellet and the complex dielectric permittivity, which were determined to be ε' = 3.01 and ε''=0.049, values typical for molecular magnet samples. In addition, five resonance lines are observed which we attribute to resonance transitions within the S=12 multiplet. Thus, the highest-frequency line is assigned to the 12→12 transition, and so on. The lines are much narrower (11 μeV FWHM) than those observed for other SMMs, e.g., 23 μeV FWHM for Mn₁₂Ac. The fit procedure showed that the lines are inhomogeneously broadened and best described by Gaussian line shapes. The small linewidth indicates that distributions in ZFS parameters (D strain) are small in these samples. A fit of the GSH parameters [Eq. (1)] to the observed resonance frequencies, yields D_{5a12} = −0.368 cm⁻¹ (0.0456 meV) and B₂ = −4.0 × 10⁻⁶ cm⁻¹ (4.96 × 10⁻⁷ meV) best parameter values. The theoretical energy barrier calculated from these ZFS parameters is U_{theo} = 76 K, which is much larger than the experimentally found U_{eff} ≈ 53 K, indicating that the molecule can shortcut the barrier in some way. The ZFS values are in themselves not remarkable and close to those reported for other manganese clusters with similar ground state spins,
e.g., $D_{s=10}=-0.457 \text{ cm}^{-1}$ for Mn$_{12}$Ac (Ref. 25) and $D_{s=17/2}=-0.247 \text{ cm}^{-1}$ for Mn$_9$,\(^\text{19}\) Interestingly, the fourth-order axial ZFS is currently accepted to parametrize effects of mixing between spin multiplets ($S$ mixing),\(^\text{42}\) which would mean that $S$ mixing is only limited, contrary to expectation. However, the fit does not simulate the resonance line positions and intensities satisfactorily, which is in contrast to the situation for other molecular nanomagnets that feature strong $S$ mixing.\(^\text{e.g., Ni}_4,\text{18,43}\) Therefore, the investigated Mn$_9$ SMM represents an example where the giant spin model cannot satisfactorily describe FDMR spectra and it will be shown below that this is due to a complete breakdown of the giant spin model. It will also be shown that the resonance line at 0.80 meV is due to a transition within the $S=11$ excited multiplet. However, removal of this resonance line does not result in a better fit. The calculated line intensities are much larger than those experimentally found, especially for the highest-frequency lines. This we attribute to a combination of parasitic radiation in the cryostat, and the presence of many more states than taken into account by the giant spin model, which decreases the relative population for any given state.

Similar FDMR results were obtained for (3) (Fig. 8) and six sharp resonance lines were observed. A fit of the GSH parameters to the observed resonance line positions yields the following values: $D=-0.362 \pm 0.001 \text{ cm}^{-1}$ ($-0.0449 \text{ meV}$) and $B'_S=-6.0 \pm 0.4 \times 10^{-6} \text{ cm}^{-1}$ ($-7.4 \times 10^{-7} \text{ meV}$). The simulated spectrum matches the experiment much more closely for (3), especially for the high-frequency lines. Interestingly, the theoretical energy barrier ($U_{\text{theor}}=75 \text{ K}$) is virtually the same as for (2) but smaller than the experimentally found energy barrier ($U_{\text{expt}}=86 \text{ K}$). This unprecedented finding means that the magnetization relaxation must involve states that do not belong to the ground-spin multiplet.\(^\text{7}\) Again, we turn to INS to determine the positions of the excited-spin multiplets, which will allow full characterization of the system.

Figures 9(a) and 9(b) show the high-resolution INS experimental data for compounds (2) and (3), respectively, collected on IN5 with an incident wavelength of 6.7 Å (53 $\mu$eV FWHM resolution at the elastic peak). At the lowest temperature $T=2 \text{ K}$ only the ground state is populated and, due to the INS selection rules, only transitions with $\Delta S=0, \pm 1$ and $\Delta M=0, \pm 1$ can be detected. The lowest-energy excitation can be thus easily attributed to the intramultiplet transition from the $|S=12, M_S=\pm 12\rangle$ ground state to the $|S=12, M_S=\pm 11\rangle$ first excited level. The position of this intramultiplet excitation is found to be at about the same energy in both compounds, i.e., $\sim 1.1 \text{ meV}$, indicating only small differences in the anisotropy of the two systems. In contrast the first intermultiplet $S=12 \rightarrow S=11$ excitation at about 1.41 meV in compound (2) is not visible in the spectra at 6.7 Å of compound (3). This can be understood looking at the data at higher-energy transfer, collected with an incident wavelength of 3.4 Å [see Figs. 10(a) and 10(b)]. Indeed the
The first intermultiplet excitation is considerably raised in energy in compound (3) with respect to compound (2), from 1.41 to 1.87 meV. This gives a direct evidence of an increase in the isotropic exchange parameters while the anisotropic parameters are approximately the same for both molecules. The INS spectra collected at a base temperature of 2 K, enabled us to directly access the whole set of intramultiplet and intermultiplet transitions allowed by the INS selection rules in both compounds. By raising the temperature to 16 K the intensity of the magnetic peaks decreases, thus confirming their magnetic origin. A total of five intermultiplet excitations for compound (2) toward different $S=11$ excited states can be detected. For compound (3) four intermultiplet excitations have been observed. All the magnetic excitations are marked in Fig. 10 with the corresponding transition energies.

To complete our investigations of the transitions within the $S=12$ ground-state multiplet, we additionally performed high-resolution measurements of molecule (3) using IN5 with incident wavelengths of 10.5 Å (FWHM=13 μeV at the elastic line) (see Fig. 11). These measurements allowed us to observe transitions originating from the top of the anisotropy barrier.

A further confirmation of the good assignment of the observed excitations is provided by the study of their $Q$ dependence. As revealed by Fig. 12, the intramultiplet transition ($\Delta S=0$) shows a distinctive $Q$ dependence, with a pronounced intensity at low $Q$, that dies out quite rapidly following the Mn$^{3+}$ form factor. In contrast, intermultiplet excitations present flatter behavior with considerably less intensity at low $Q$. The assignment of the observed excitations to intramultiplet or intermultiplet transitions has been confirmed by comparison with FDMR measurements performed on both compounds (see Figs. 7 and 8). The position
of the intramultiplet INS transitions are consistent with the
FDMR measurements performed on the same sample (see
Table I). Due to the different selection rules of INS and
FDMR, we can conclude that all the peaks observed at T
=2 K above 1.2 meV energy transfer correspond to inter-
multiplet transitions since they are absent in the FDMR
spectra.

The straightforward assignment of the base temperature
observed excitations allows us to draw some considerations
on the experimentally deduced energy-level diagram. For
both compounds, a rough estimate of the splitting of the spin
ground multiplet gives \( |D|S^2 = 6.5 \) meV. This value is com-
parable to the energy interval explored by the high-energy
transfer INS data (Fig. 10), where most of the intermultiplet
\( S=12 \rightarrow S=11 \) excitations have been observed. This
experimental observation leads to the conclusion that also in (2)
and (3) several excited states lie within the anisotropy split
ground state with the consequent breakdown of the GSH
approximation. Due to the inadequacy of the GSH for (2)
and (3), the microscopic spin Hamiltonian [Eq. (2)] was used
to model the data and extract the exchange constants and
anisotropies. The minimal set of free parameters is given by
three different exchange constants \( J_{11} = J_1, J_{12} = J_{23} = J_{13}
= J_{12} = J_{23} = J_{13} = J_2 \), and \( J_{13} = J_{123} = J_3 \) (Fig. 6) and two
sets of CF parameters \( d_1 = d_1, c_1 = c_1, \) and \( d_2 = d_2, c_2 = c_2 \).
Indeed, the ligand cages of sites 1 and 3 are rather similar
and we assumed the corresponding CF parameters to be
equal. Since experimental information is insufficient to fix
independently the two small \( c \) parameters, we have chosen to
constrain the ratio \( c_1/c_2 \) to the ratio \( d_1/d_2 \).

The isotropic exchange and crystal-field parameters de-
duced by the simultaneous best fit of the experimental data
are reported in Table II. Figure 13 shows the calculated
energy-level diagram using the best fit procedure for Eq. (2)
(left) and the GSH model (right) for (2) and (3).

### IV. DISCUSSION

The experimental data collected on the three variants of
\( \text{Mn}_6 \) clusters provide direct evidence that a general feature
for this class of compounds is the nesting of excited mul-
tiplets within the ground-state multiplet. This is an unavoid-
able effect when the isotropic exchange parameters have the
same order of magnitude as the single ion anisotropy param-
eters, as it happens to be for \( \text{Mn}_6 \). The nesting of spin states
can be clarified by observing the energy-level diagrams for
the three molecules presented in Figs. 5 and 13. The diagram
on the left shows the energy levels calculated by a diagonal-
ization of the full spin Hamiltonian while the energy-level
scheme on the right-hand side has been calculated consider-
ing the GSH approximation. It is clear that the GSH does not
account for any of the spin states with \( S \) different from \( S_{GS}
\) that lie within the split GS energy-level diagram. The above
states represent a shortcut for the relaxation of the magneti-
zation and can promote resonant intermultiplet tunneling
processes that manifest as additional steps in the magnetiza-
tion curve absent in the GS model.\(^8,41,44-46\) The overall result
is a lowering of the effective anisotropy barrier with respect
to an ideal molecule where the spin ground state is well
separated from the excited ones, as was first demonstrated in
Ref. 7.

| Table II. Isotropic exchange and CF parameters for Eq. (2)
in meV deduced by fitting INS and FDMR data for the two \( \text{Mn}_6 \)
clusters with \( S=12 \). |
<table>
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<tbody>
<tr>
<td>( U_{\text{eff}} ) (K)</td>
<td>( J_1 )</td>
<td>( J_2 )</td>
<td>( J_3 )</td>
<td>( d_1 )</td>
<td>( d_2 )</td>
</tr>
<tr>
<td>(2) 53</td>
<td>-0.61</td>
<td>-0.31</td>
<td>0.07</td>
<td>-0.23</td>
<td>-0.97</td>
</tr>
<tr>
<td>(3) 86.4</td>
<td>-0.84</td>
<td>-0.59</td>
<td>0.01</td>
<td>-0.20</td>
<td>-0.76</td>
</tr>
</tbody>
</table>

\(^{a}\)Not observed.
The projection coefficients $a_i$ of the single ion anisotropy to spin states of different $S$ values can differ significantly, giving rise to considerably different $D$ values. The ligand field study of various members of the Mn$_6$ family (Ref. 48) provides experimental evidence of this. Recent theoretical studies proposed that the intrinsic relationship between $S$ and $D$ causes a scaling of $U$ that goes approximately with $S^0$ (see Refs. 4 and 28), raising the question whether it is worth trying to increase the value of spin ground state to obtain a larger energy barrier. Indeed, higher spin ground states would correspond to lower $D$ parameters, neutralizing the overall effect on the height of the anisotropy barrier. In recently performed electron paramagnetic resonance studies the authors proposed that the barrier goes roughly with $S^1$ instead. In the specific case of Mn$_6$, because of the very large $S$ mixing, the projection onto a well-defined spin state is no more justified and it is not possible to associate the barrier $U$ to a defined $S$ value. However, if we consider the effective anisotropy barrier for artificially isolated $S=4$ and $S=12$ states [i.e., $U=47$ K for (1) and $U=105$ K for (2)], we can confirm that the barrier does not go quadratically with $S$, as one could naively deduce from the equation $U = |D|S^2$. Indeed, $U_{S=12}/U_{S=4} \approx 2.2 \approx 12^2/4^2 = 9$. This confirms what has been pointed out in Ref. 4, i.e., even though the highest anisotropy barrier is obtained with the molecule with the highest spin ground state, the increase in the total spin is not as efficient as one would expect and alternative routes, such as increasing the single ion anisotropy, should be considered.

V. CONCLUSION

We have performed INS and FDMR measurements on three variants of Mn$_6$ molecular nanomagnets, which have
the same magnetic core and differ by slight changes in the organic ligands. INS measurements have unambiguously evidenced the presence of low-lying excited states in all the three molecules. The combination of the two techniques enabled us to determine the spin Hamiltonian parameters used for the analysis of the magnetic properties. The nesting of excited states within the ground-state multiplet strongly influences the relaxation behavior and plays a crucial role in lowering the effective energy barrier. The calculations of the relaxation dynamics give results that are consistent with the experimental values and show that the highest barrier is obtained for ideal molecules with an isolated ground state. This observation might be valid for a wider class of SMMs and suggests that the combination of a high uniaxial anisotropy together with strong intramolecular exchange interactions is necessary to hinder the relaxation of the magnetization and engineer molecules able to retain the magnetization for a long time.

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