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Offshore conversion of wind power to gaseous fuels: Feasibility study in a depleted gas field

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Abstract
A proof-of-concept study is presented of a Power-to-Gas system that is located fully offshore. This paper analyses how such a system would perform if based at the depleted Kinsale Gas Field in the Celtic Sea Basin off the south coast of Ireland. An offshore wind farm is proposed as the power source for the system. Several conversion technologies are examined in detail in terms of resource efficiency, technological maturity, and platform area footprint, the aim being to ascertain their overall applicability to an offshore Power-to-Gas system. The technologies include proton exchange membrane electrolysers for electrolysis of water to release H2. Bipolar membrane electro-dialysis and electronic cation exchange module processes are also considered for the extraction of CO2 from seawater. These technologies provide the feedstock for the Sabatier process for the production of CH4 from H2 and CO2. Simulations of the end-to-end systems were carried out using Simulink, and it was found that the conversion of offshore wind power to hydrogen or methane is a technically feasible option. Hydrogen production is much closer to market viability than methane production, but production costs are too high and conversion efficiencies too low in both cases with present-day technology to be competitive with current wholesale natural gas prices.

Keywords
Energy storage, electrolysis, Sabatier process, methanation, syngas, CO2 extraction from seawater

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Introduction
This paper addresses three interconnected issues: the re-use of end-of-life oil and gas platforms, carbon dioxide capture and re-use, and the storage of marine renewable energy. By 2010, at least 7668 offshore platforms had been installed in ocean waters worldwide,1 where on average each platform represents 1000–20,000t of steel.2 As these structures come to the end of their design lives and offshore hydrocarbon fields cease to be exploited, many offshore platforms are being decommissioned, even abandoned. For example, Wan Abdullah Zawawi et al.3 have reported that 48% of platforms in Malaysian waters alone have exceeded their expected 25-year lifetime. The European Commission Directive 2008/98/EC on waste outlines the following hierarchy: Prevention, Re-use, Recycling, Recovery, Disposal.3 Whilst recycling steel is of legitimate benefit (Wan Abdullah Zawawi et al.2 estimate that the 475 Mt of steel recycled in 2008 would mitigate 811 Mt in CO2 emissions in producing hot rolled steel), re-use is clearly preferable. Repurposing these platforms for ocean colonisation has been suggested; rigs-to-reef programmes exist for platforms in shallow waters in order to enhance offshore habitats, but are ill-suited to deeper waters4 and provoke strong opposition from environmentalists.2 It is therefore opportune to ask whether such platforms could be reutilised as part of a more sustainable energy future.
Anthropogenic CO₂ emissions are a continuing global concern. The Intergovernmental Panel on Climate Change has recommended that such emissions be cut by 30–85% by 2050 so as to return the CO₂ concentration within the atmosphere to 350–440 ppm. To achieve this, it may also prove necessary to remove excess CO₂ already absorbed by the terrestrial and marine environment. Doney et al. suggest that about one-third of anthropogenic carbon released to the atmosphere later becomes dissolved in the ocean – this is a mitigating factor for climate change to an extent, but causes ocean pH levels to drop, and the fundamental chemical balance to alter, leading to ocean acidification (termed ‘the other CO₂ problem’).

Global incentives to develop clean, renewable energy sources, have given rise to the concept of Power-to-Gas (PtG), which involves conversion of electricity, typically harnessed from a renewable source, to a gaseous energy carrier such as hydrogen or methane. Where methane is the desired end product, the opportunity for CO₂ removal presents itself, such as the PtG process proposed in this paper which utilises CO₂ taken from directly from the ocean. Thus, this paper investigates the possibility of implementing an offshore PtG system operating exclusively on renewable resources at an existing gas platform. The system will allow for the re-use of end-of-life gas infrastructure and provide a means of converting variable offshore wind power into an always-available energy format.

PtG

PtG is the functional description given to the conversion of electrical power into a gaseous energy carrier. The drive towards making PtG a feasible option in the global energy mix is in many ways linked to currently perceived shortcomings of the renewable energy sector. Intermittency of supply and curtailment due to mismatch with demand and grid limitations adversely affects the value of renewable energy sources such as wind. This fluctuating curtailed energy can be stored by converting the electricity generated into hydrogen or methane. The stored energy can be reconverted to electricity when required through standard power generation, or the gas can be used directly for heating or transport.

The increasing proportion of electricity derived from fluctuating, non-synchronous generation sources such as wind turbines is forcing grid operators to curtail output of wind farms at certain times. It has been estimated that in the absence of mitigation measures up to 7% of wind energy production in Ireland would have been curtailed by 2020. Previous studies have shown the potential of using wind-generated electricity that would otherwise be curtailed to generate hydrogen to upgrade biogas derived from anaerobic digestion reactors. The production of hydrogen and methane in this proposed offshore catalytic based PtG system is described in ‘Hydrogen production’ and ‘CO₂ extraction and renewable methane production’ sections.

Hydrogen production

One prospect of a sustainable future is offered by the so-called hydrogen economy based upon the simple chemical reaction between hydrogen and oxygen molecules whereby heat and electric energy (in a fuel cell) are released and water is the sole by-product. It has been estimated that 1 kg of hydrogen is the equivalent of one US gallon of gasoline with regard to energy provided. However, molecular hydrogen is not abundant in the atmosphere, with a concentration of only 0.00005% in air; instead, hydrogen is normally bound into more complex molecules such as water and hydrocarbons. Although hydrogen could be extracted from hydrocarbon compounds using fuel-processing technologies, such technologies are non-sustainable due to the use of fossil fuels during processing and the use of fossil fuels as feedstock. An alternative technology for hydrogen production is water electrolysis, which uses electricity to split water into hydrogen and oxygen. Given the availability of renewable electricity, electrolysis has been identified as a key component for PtG. Electrolysis essentially involves application of direct electric current to water, to split it into its constituent components hydrogen and oxygen. As of 2013, electrolysis provided ~4% of global hydrogen demand. The overall chemical reaction during electrolysis is as follows

\[ 2\text{H}_2\text{O}(l) \rightarrow 2\text{H}_2(g) + \text{O}_2(g) \]

Alkaline electrolysis cells (AECs), solid oxide electrolysis cells (SOECs), and proton exchange membrane (PEM) electrolyser are three examples of electrolysis technologies. SOECs are still in the development stage, whereas AECs previously appeared to have been limited in their applicability to PtG systems owing to their reportedly slow cold-start deployment time, which could impact on their ability to handle fluctuating electricity supplies (typical of renewables such as wind, wave, or photovoltaics). However, Grond et al. report that this shortcoming is merely a result of lack of demand for flexible operation, and AECs are fully capable of flexible operation with ramp-up times of seconds over a power load range of 5–100%. The largest onshore PtG plant to date, developed by ETGAS for Audi AG, has a 6.3 MW el capacity and utilises high-pressure AECs with an electrical load range of 10–110%, and a ramp-up rate of 15% per second, and 0–100% per 5 min. It should be noted that the corrosive nature of the alkaline electrolyte used in AECs has led to concerns regarding sustainability issues. The third
technology, PEM electrolysers, utilises very thin (μm-scale) proton conductive polymer electrolyte, which facilitates rapid response times to intermittent energy sources such as renewables. PEM electrolysers have efficiencies similar to those of AECs of 67–82% and it is expected that PEM electrolyser efficiencies will improve in the coming years.13 ITM Power and Shell have recently announced the planned installation of a 10 MW PEM electrolyser at Shell’s Rhinelan refinery plant, and to use the hydrogen produced on site for hydrocracking fossil fuel as well as the provision of grid services. Further details on each of the foregoing electrolysis processes are given in Grond et al.13 PEM electrolysers with their perceived good efficiencies, high maximum current densities suited to constrained locations such as offshore rigs,15 modularity and flexible operation abilities were selected for the PtG systems considered herein.

The feed water used in electrolysis must be of high purity and satisfy a maximum allowable limit of 0.5 ppm total dissolved solids (TDS),16 whereas seawater in the Atlantic Ocean has a TDS concentration of 38,500–40,000 ppm.17 Reverse osmosis (RO) and deionisation post treatment are pre-requisites for seawater electrolysis systems in order to achieve required water conditions (American Society for Testing and Materials Type II). Pre-treated seawater is pumped at high pressure into the RO module. A series of semi-permeable, thin-film composite membranes then separates the stream into pure water permeate and a salt-retaining concentrate. Osmotic pressure is the absolute minimum pressure required for the system to operate at its thermodynamic limit.18 Typical purified water production rates are 45% for a single pass and can be theoretically increased to 85–90% by introduction of a second pass. However, due to osmotic pressure limitations, realistic purified water extraction is capped at about 60%.19

**CO₂ extraction and renewable methane production**

The source of the carbon dioxide is a major factor in the process of renewable methane production as is the energy required to capture and deliver the carbon dioxide to the methanation unit. Where the desired end product is methane, both hydrogen and carbon dioxide need to be sourced.

For a proposed offshore PtG platform, one approach is to utilise existing adjacent resources in order to offset the increased risk and cost incurred because of location in the marine environment, noting that the ocean has a CO₂ concentration about 175 times that of air.20 Several candidate processes are presently under consideration. One involves extracting carbon dioxide from seawater using bipolar membrane electro-dialysis (BPMED), which has reportedly a total CO₂ recovery efficiency of 59%.21 Another alternative under development by the U.S. Naval Research Laboratory (NRL) comprises a novel three-chambered electronic cation exchange module (ECEM) approach which has been demonstrated in a marine environment at a small scale (1900 ml/min seawater feed-rate). One advantage of ECEM over the BPMED process is the simultaneous production of both carbon dioxide and hydrogen, should methane be the desired end product. The ECEM process consists of influent seawater being acidified by the strong exchange of cations in the electrode compartment, thus aiding the extraction of CO₂ bound as carbonate and bicarbonate. Laboratory tests conducted by the U.S. NRL indicate a total extraction rate of 92% from natural seawater at pH ≤ 4.20 Reported production rates of CO₂ and H₂ are 0.003–0.004 and 0.0093 mol/min, respectively, for a flow rate of 1900 ml/min of seawater.20 Further supplies of H₂ would be required to satisfy the methanation process as per the Sabatier Equation used in this study, which requires four times more hydrogen than carbon dioxide

\[
\text{CO}_2 + 4\text{H}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O}
\]

The ratio of hydrogen to carbon dioxide produced by the ECEM would also be suitable for methanol production which requires three times more H₂ than CO₂. This is another promising energy transformation but is not dealt with in this study. The capturing and storing of the additional CO₂ produced by the ECEM above what is required for the Sabatier methanation process has been investigated in this study.

The NRL research to date has focused on the development of the ECEM at proof-of-concept level through modifying an off-the-shelf electro-deionisation unit to function as an ECEM. Willauer et al.20 state that no attempt was made to make the ECEM (585 kW h/kg H₂ at STP) as energy efficient as a typical commercial water electrolysis unit (51 kW h/kg H₂ at STP).

**Offshore storage and transportation**

For an offshore PtG plant to be feasible, infrastructure must be in place to accommodate storage and transport of the gaseous fuels produced. Where H₂ as an end product is concerned, pressure tanks (presently 4–400 bar, with 700 bar storage under development) are usually incorporated, with 88% of operating onshore PtG plants opting for this form of storage as of 2013.22 Although an inherent advantage of high pressure storage is the reduction in required space as storage pressure is increased, the overall system efficiency may drop due to the need for a hydrogen compressor.22 This poses an interesting question as to whether to situate such containers onshore or on bespoke floating structures offshore. The latter option would require more stringent design criteria associated with the harsher
environmental conditions offshore, but these factors may be offset by a decrease in installation time and cost. This leads to a novel idea; there exists a window of opportunity to retrofit abandoned rigs from depleted gas reservoirs, and repurposing them as offshore PtG farms. The gas storage system requirements will be determined by the existing infrastructure available in terms of existing pipe lines ashore, the depleted gas reservoirs, and the level of renewable gas production at a site.

A second gas storage option is to make use of existing geological formations, as proposed for carbon sequestration. Noting again the possibility of utilising existing infrastructure, the locations of porous rock structures left over from depleted oil and gas fields are obviously well known.

Submerged isobaric energy bags offer another novel approach to offshore gas storage and are currently being investigated in the context of compressed air energy storage. However, such energy bags are designed to operate in water depths of the order of several hundred metres, and so it may not be feasible to co-locate them together with shallower depth fixed and floating foundation offshore wind energy converters.

Logistical and economic issues must be addressed with regard to transporting the end products either by pipeline or by ship for onward distribution. For pipelines, the inherent installation costs might be avoided by co-locating PtG farms at sites of depleted gas fields where connections to the gas grid are already in place. Under such circumstances, the limiting factor for a power-to-hydrogen (PtH) system would be dictated by the maximum allowable hydrogen concentration in the gas mixture. For example, in Germany, hydrogen is currently limited to 5% by volume to ensure minimum quality within the grid.

The gas product transportation format explored in this study is the use of the existing gas pipe lines to shore connection to the natural gas (NG) grid. Given that the export capacity of the pipe lines is several times larger than the renewable gas peak production, storage on the platforms is not considered in the analysis in this study (see ‘Site description’ section for further details). An advantage of this particular type of site and gas storage format is the connection to the gas grid which allows gaseous products to be dynamically produced and directly exported, avoiding the need for onsite H2 storage which was found by Götz et al. to contribute up to 21% of capital costs of a PtG plant.

**Case study**

The Old Head of Kinsale Gas Field was selected as a hypothetical location for the offshore PtG concept. This site is nearing the end of its gas production lifespan and has suitable infrastructure in place for conversion to PtG use. An overview of the scenarios analysed in this paper is given in Table 1, with a detailed description of the simulation methodologies in ‘Simulation methodology and PtG scenarios’ section.

**Site description**

Figure 1 shows the location of the Kinsale Gas Field in the Celtic Sea, approximately 50 km south off the coast of Co. Cork, Ireland. The gas field was first discovered in 1971, with several further satellite sites discovered over the next three decades. The gas field is located in a water depth of 90 m, with reservoirs > 900 m below the seabed. Two production platforms constructed of steel, Alpha and Bravo, were installed in 1977 to facilitate the extraction of NG.

A gas storage facility is currently operated by PSE Kinsale Energy Ltd (KEL) at the main site. The facility has a working volume of 230 million standard cubic meters, which is equivalent to 2472 GW h, or ~5% of Ireland’s annual gas consumption in 2013/2014. The maximum withdrawal rate is 29.3 GW h/d, and maximum injection rate into the gas grid is 27.6 GW h/d. As gas production has begun to wind down, KEL has indicated that existing storage operations are not economically viable in the long term without further development. Upgrade of the storage facility to a PtG terminal may therefore be worth consideration, perhaps on a pilot basis. Gas storage operations ceased in 2017 and production is anticipated to completely cease by 2021 with plans for decommissioning shortly afterwards.

**Existing infrastructure**

The existing two steel platforms could be reconfigured for PtG production, to house either conversion apparatus or else storage containers for gaseous end products. The Alpha platform has deck dimensions of 53 m x 25 m and an elevation of 122 m above the seabed. It is supported by eight 1.56 m diameter legs, braced by I-beams and raking tubular members. The Bravo platform has similar dimensions.

Ireland has a potential total offshore storage capacity of 94,000 Mt of CO2 with the geological formations in the Celtic Sea Basin particularly suitable for gas storage. The depleted gas fields of the Kinsale Gas Field offer a storage potential of some 330 Mt of CO2. However, such a venture would have concomitant risks associated with geological storage, in particular containment risks due to the shallow location of the aquifers and the possibility that geological seals may have become compromised by pressure depletion exacerbated by the presence of several faults cutting the top of the reservoir.

**Wind resource assessment**

Hourly wind speed data were obtained from the Irish Marine Institute’s website (www.marine.ie) and
checked for continuity. The data were recorded at one of the Kinsale platforms between 2003 and 2008 at an elevation of 66 m above mean sea level. The year 2004 was chosen from the measurement archive for subsequent analysis because it has the most complete data series. The chosen year was also considered representative of the average wind climate at the site. Our analysis of longer-term records from the nearby Cork Airport meteorological station indicates that the 2004 mean wind speed was equivalent to 98% of the long-term mean. Wind speeds were extrapolated to the proposed turbine hub height of 100 m above sea level (asl) using the log law, assuming neutral atmospheric stability and a constant surface roughness length of 0.001 m giving an annual average wind speed of 9.4 m/s.

Turbine selection

It has recently been estimated using levelised cost of energy calculations that a floating platform can be economically competitive with a fixed structure under optimised conditions. Examples of deep water, floating wind farms include Statoil’s Buchan Deep Hywind pilot project (30 MW) and Hexicon’s cancelled Dounreay Tri project (10 MW), both off the Scottish coast. Taking account of the foregoing, we decided to consider a floating offshore wind farm comprising of twelve 6 MW Siemens D6 turbines individually mounted to Hywind II type floating foundations.

Simulation methodology and PtG scenarios

Electricity generation simulations were carried out using MATLAB Simulink R2013b (MathWorks, USA) using hourly average wind speed values. Three different PtG technology options are considered for the direct conversion of wind-generated electricity to gas at the Kinsale site in Scenarios 2–4. For each time step, the Simulink model converts wind speed into electrical power.
Based on prescribed power curves and the number of turbines at the wind farm.

Scenario 1 (OWF) comprises an offshore floating wind farm with new subsea cables ashore, whilst Scenarios 2–4 are PtG systems with hydrogen and methane gases being injected into the existing NG infrastructure at the site. Scenario 1 provides a reference electricity-only state with which to compare the results of the PtG gas scenarios.

Scenario 2 (PtH) sees hydrogen gas produced by PEM electrolyzers with feed water being supplied by RO of seawater, the hydrogen is injected directly into the existing gas grid on site.

In Scenario 3 (ECEM + PEM), the ECEM plant produces CO₂ and H₂ from seawater. The PEM electrolyzers are used to generate additional hydrogen, as the H₂:CO₂ ratio from ECEM is not sufficient to convert all the available CO₂ to methane. Methanation is carried out using a Sabatier reactor, and methane is injected into the gas grid on site.

Finally, Scenario 4 (ECEM + CO₂ seq) is similar to Scenario 3, with the omission of the PEM electrolysis and associated RO plant. Methanation is carried out using a Sabatier reactor and methane is injected into the NG grid. The surplus CO₂ generated by the ECEM process is sequestered.

Quantities of electrolyser feed water required and associated energy for pre-treatment of the seawater are calculated at each time step, depending on the power available and the rate of operation of the electrolyser plant in Scenarios 2 and 3. The energy required to purify and pump the feed water is deducted from the power available for gas production.

Table 2 outlines the efficiencies of the individual conversion processes. The three PtG scenarios are graphically illustrated in Figure 2(a) to (c).

Capacity and plant sizing parameters were initially set by the space available on the existing platforms. In all cases, the rate of gas production was far less than the capacity limit of the existing gas production infrastructure. Given the dimensions of the existing platforms at the Kinsale site, it is envisaged that 70 shipping containers could be accommodated, corresponding to c. 70 MW of electrolyser, as discussed in ‘Hydrogen production’ and ‘Results’ sections. This space limit fixes the size and rated capacity of all equipment used in Scenarios 1–4.

The cost of gas was determined using the formula

\[
\text{Gas Cost per unit} = \frac{\text{Equivalent Annual Cost} + \text{O&M}}{\text{Annual Gas Production}}
\]

(1)

\[
\text{Equivalent Annual Cost} = \frac{\text{CC} \times r}{1 - (1 + r)^n}
\]

(2)

where CC is the total capital cost (component replacement costs included), r is the discount rate, n is the time period, and O&M is the annual operation and maintenance cost.

**Results**

The mean wind speed at an elevation 100 m asl is calculated to be 9.4 m/s, as discussed in ‘Wind resource assessment’ section. The gross turbine capacity factor was 58%, equivalent to 5072 full-load hours per annum. A single Siemens D6 turbine operating under these conditions would generate 30.4 GW h of electricity over a one-year period.

In order to assess the feasibility of the proposed systems, the practical PtG farm deployable at the existing Kinsale Gas Field platforms was considered in detail. A limiting factor with regard to sizing the farm is the number of electrolysers to be housed, with half of each platform being allocated for electrolysis. Here, 1 MW electrolysers were selected, each made up of four 250 kW units housed in a standard shipping container (6.1 m × 2.44 m × 2.59 m). Assuming that the platform can be stripped down so that only the deck remains, this permits 35 electrolysers to be housed on each platform, allowing 2 m spacing between rows. Mobile RO units are situated on each platform, and based upon the Zeppelin model for

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**ASTM**: American Society for Testing and Material; **ECEM**: electronic cation exchange module; **PEM**: proton exchange membrane; **RO**: reverse osmosis.
quantifying spatial requirements. In order to produce up to 12.5 m$^3$/h of desalinated water, 12 pre-treatment units (six on each platform) are placed in standard shipping containers, in order to meet the electrolysis requirements of the energy pathways (outlined in Table 3). The remainder of each platform is assumed to have sufficient space left to accommodate the Sabatier reactor, noting that it was not possible to quantify the reactor’s spatial dimensions, nor those of the ECEM system used in Scenarios 3 and 4. The ECEM unit incorporates RO, and so there are reduced RO system requirements for this option when compared to electrolysis. This design leaves unused space on each platform for the PtH option; however, in order to present a fair comparison of the three production options, the farm size is not increased to maximise the use of this leftover space.

Table 3 provides details of the gas production for each of the energy pathways. The overall efficiency of each system is defined as the final energetic value of the gaseous fuel produced as a percentage of the energy output from the wind turbines

$$\eta = \frac{\sum \text{Final gaseous fuel energetic value}}{\text{Wind power}} \times 100$$

(3)

These production figures are supplemented by costs incurred during the process. An economic analysis was facilitated by the Simulink model, with input parameters listed in Table 4.

Wind farm development costs are taken from International Energy Agency Energy Technology Systems Analysis Programme and International Renewable Energy Agency.$^{34}$ Capital costs are extracted from the upper bound of those presented in order to reflect the increased capital investment associated with the water depth and distance to shore. A 30% reduction in total wind farm capital costs was then applied to account for the elimination of cabling and balance of plant requirements for Scenarios 2–4,$^{34}$ where all power will be converted to gaseous fuel on site. Electrolyser costs are
estimated from values reported by ITM Power. Costs for the methanation process are based upon a combined electrolyser and methanation cost of €2000/kW. These are consistent with the methanation costs of €300–500/kW reported in Lehner et al. Estimates of the cost of ECEM units are assumed to be €1.5 million/MW as per electrolysis unit costings. For the RO cost, the lower bound is determined from figures given by Ghaffour et al., and converted at a rate of €1 = $1.105, so that the proposed system is given an optimistic appraisal of its economic feasibility. Table 5 presents the forecast production costs and end-to-end efficiencies (equation (3)) for each energy pathway.

### Discussion

As can be seen from the preceding results, none of the three scenarios produces gas at a cost comparable to present NG wholesale market prices (~€0.02/kW h). The PtH case fares the best, and the costs fall within the range of most of the studies reviewed by Götz et al. Hydrogen can also be injected directly into the gas grid upon production. When the production and flow of NG from the reservoir is discontinued as planned, the PtH scenario would be transporting 100% hydrogen through the existing steel subsea pipelines raising leakage and hydrogen embrittlement concerns. When injecting the H2 to the onshore NG grid, the H2 will be blended with NG, and blend ratios are currently limited, with a 10% per volume upper limit advised by Altfeld and Pinchbeck and 15% cited by NREL.

The electricity-to-methane pathway incorporating ECEM is the worst from an economic perspective, as might be expected, given that the process is still at a pilot stage of development and the conversion efficiencies are low. For Scenario 4, the excess CO2 produced by the process (1748 t/a) is sequestered and if a carbon cost of €100/t is applied the cost of methane production reduces from €2.21 to €2.20 per kW h. Alternative transformation pathways such as power-to-methanol or power-to-ammonia may prove to be more economically attractive, but these may incur additional costs for storage and offloading.

If the ECEM process as applied in this study were to proceed to a commercial stage, it is anticipated that the process efficiency would improve from 5.3% and gas production costs would therefore decrease. The prospect of coupling renewable fuel production with the removal of CO2 from the atmosphere via the oceans would also be attractive from a sustainability perspective.

It is obvious that the opportunity to repurpose the existing Kinsale platforms to act as an offshore PtG farm is not economically feasible at present. This finding may also hold for offshore platforms globally due to the general applicability of present methodology adopted herein. Moreover, the case would have exploited the abundant wind resources available off the south coast of Ireland.

Our Simulink model was based upon process efficiencies as opposed to experimental results; therefore, it was not possible to examine how sensitive the gas
production costs are to wind farm capacity. Additional electrolysers could perhaps be incorporated onto existing platforms through stacking of containers, subject to design checks. The carrying capacity of each platform jacket foundation is approximately 4000 t.

It could also prove useful to compare these projected costs with those involved in decommissioning. Detailed estimates of the latter are not widely available; however, based on decommissioning estimates published by Oil & Gas UK\textsuperscript{42} of £4000/t (topside) and £4500/t (substructure), the c. 8000 t Kinsale platforms could cost c. €38m each to fully decommission (assuming a conversion rate of 1€ = 1.14£).

The capital costs associated with methanation equipment remain a source of uncertainty. However, the estimate of €500/kW capacity used in this study is broadly consistent with other estimates in the literature. For example, Saric et al.\textsuperscript{43} estimated a capital cost for methanation of €286/kW and DNV KEMA’s 2013 report estimated current costs at €700/kW for kW-scale units and suggested that costs would decrease to €300–€500/kW at greater scales.\textsuperscript{13}

The cost of offshore wind continues to fall. In the UK’s September 2017 allocation round for contracts for difference projects, the lowest strike price for offshore wind was £57.50/MW h (2012 UK prices; £57.50 was approximately €64.00 as of September 2017).\textsuperscript{44} By examining the long-term feasibility of offshore PtG, it has been suggested that offshore wind capital costs will fall from ~€3 million/MW in 2012 to €1.2 million/MW by 2030,\textsuperscript{45,46} and that PEM electrolysis might attain efficiencies up to 93%.\textsuperscript{13} Without accounting for any other cost reductions, or decreases in O&M costs, these improvements would lead to a 40% reduction in hydrogen gas production costs to ~€0.09/kW h for the PtH option. Given that NG is a fossil fuel, with finite reserves, it is therefore possible that offshore PtG or more likely PtH could become a feasible energy source in the future.

**Conclusion**

This paper examined the prospect of repurposing existing, offshore gas platforms as PtG farms. The study considered three power to gas options, PtH, PtG, and ECEM at a site off the southwest coast of Ireland. It was found that the technology exists for a complete, end-to-end conversion process, entirely transforming renewable electricity from fluctuating sources, such as wind or wave power, to readily stored hydrogen or methane. In the case of ECEM, CO\textsubscript{2} is also sequestered from the ocean. Using present-day technology with its attendant costs, the repurposing of gas production platforms does not yet appear to be economically viable, but PtH technology is far closer market viability than PtG using ECEM. The PtH scenario produced hydrogen at €0.16/kW h, some eight times the wholesale market cost of NG, whilst ECEM and PEM produced methane at €1.54/kW h. This is largely due to the high capital costs of the technology for conversion of electricity to the final gaseous product and, in the case of the ECEM process, low conversion efficiencies. For the most viable option (electricity to hydrogen, PtH) the overall end-to-end efficiency is ~75%. This is comparable to many other utility-scale energy conversion processes (e.g. pumped hydropower round-trip efficiency is typically 80%\textsuperscript{47}). The major disadvantage is simply that electricity is a more valuable energy carrier than gas. However, forecast reductions in the capital cost of offshore wind farms and electrolysers should lower the cost of renewable gas, i.e. hydrogen, to within a factor of five of the cost of NG. As conventional NG wells decline and land to house renewable generation becomes less available, and the potential of gaseous fuels grows within the transport industry, it is possible to envisage a future where offshore PtG systems become a reality.

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